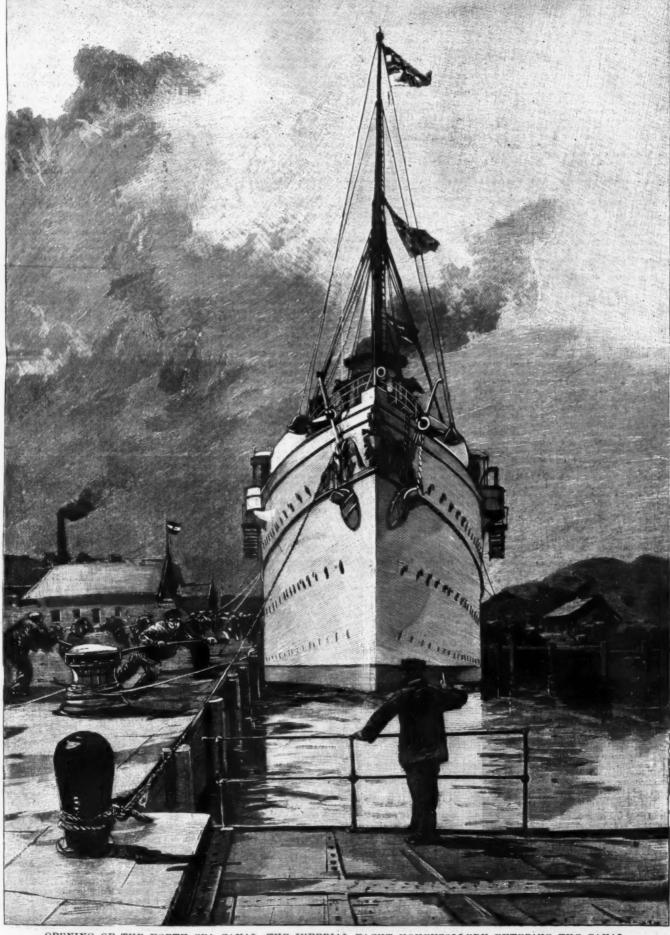


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OPENING OF THE NORTH SEA CANAL-THE IMPERIAL YACHT HOHENZOLLERN ENTERING THE CANAL

THE NORTH SEA CANAL

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The North Sea and Baltic Canal, after eight years of hard work and the expenditure of nearly eight millions sterling, has been completed and opened for traffic, and the Danish peninsula has become an island. The canal, which runs from near Brunsbuttel, in the mouth of the Eibe, to the Fjord of Kiel, on the Baltic, has a length of about sixty miles, a normal width at the surface of 197 ft. a normal width at the bottom of a little over 72 ft. and a depth over the central portion of 29 ft. 6 in., so that the largest vessels in existence, inclusive of all men-of-war, can traverse it without having to lighten themselves, and that vessels not of the largest size can safely pass one another without being obliged to lie up, or even to stop. For the convenience of the very few ships which cannot pass one another in that manner, there are, at intervals, broadenings, or "bays." where the bank to bank width is, for a distance of 820 ft., increased to 328 ft. and the bottom width to 197 ft. Nor are there any overhead obstructions. The canal is crossed by four lines of railway, two of which have comparatively little traffic. For the service of these two swing bridges, pivoted close to the bank, are provided. At ordinary times the bridges are swung round so as to lie parallel with the waterway; and only when trains are due are they thrown across to the opposite side. The whole operation can be accomplished in two minutes. Each bridge is double, a separate swing span carrying each set of rails; and thus, should any unforeseen accident occur to temporarily prevent the working of one span, the entire traffic can be conducted by means of the other. Upon the remaining two railway lines the traffic is heavier, and, in consequence, fixed bridges have been built to carry it. But these bridges are so lofty that no masted ship now affout need, before attempting to pass under them, send down more than her topgallant masts, which are spars such as, of course, she would not need in a canal in any conceivable circumst

of 138 ft. and the spans, which are the broadest in Germany, have a reach of 511 ft. A third swing bridge, at Rendsburg, is for the use of vehicles and foot passengers.

The new canal is not lockless, although it is nearly so. The Baltic Sea is almost without tides; and, therefore, a lock at the Baltic end is rendered necessary only by the possibility that occasionally a strong northeasterly gale may pile up the water to an inconvenient extent in Kiel Bay. The effect of the eastern lock, when closed, will be to prevent a special tide of that kind from flooding the banks of the canal; but ordinarily that lock will be left open throughout the twenty-four hours. At the Elbe, or western end, the situation is different. There the rise and fall are considerable, and consequently, at every ebb, in order to maintain the desired depth of water the gates must be closed for some hours, though ships will still be able to pass through, subject to the delays which are attendant upon the passage of all locked canals. The two terminal locks are very important engineering works. Each really consists of a pair of locks, placed side by side, and parallel one with the other; and the dimensions of each lock are: clear length 492 ft.; width 82 ft.; depth 33 ft. There are, so far, no ships 83 ft. broad; but there are a few ships more than 492 ft. long. Even these monsters, however, can traverse the canal whenever it is not necessary to use the locks, or, in other words, during by far the greater portion of every day. Indeed, compared with the Suez Canal, the new waterway may be regarded as a fine broad highroad, whereas the older work is but a well-kept country lane. There are many ships that cannot use the one; there are none that cannot use the other.

The advantages to the commerce of the Bitte must be considerable. On the route from London to St. Petersburg the canal saves 338 miles; on the route from Hamburg to St. Petersburg it saves 424 miles; and, since it has been estimated that there are now annually about 1,500 ships havin

plant, with the assistance of which it does not despair of keeping the way altogether open during all but the severest seasons.

The quantity of traffic that can be dealt with is almost unlimited. Work may go on night and day, seeing that the canal is lighted electrically from end to end, that there are plenty of powerful tugs at the disposal of the administration, that at each end there are capacious basins in which vessels may await, if necessary, their turns, and that a speed of 5-3 knots, or upward of six statute miles an hour, will be permitted even to merchantmen using the route. Warships will, of course, adopt such speed as service considerations may render desirable or possible; and with a view to this sort of express traffic, the banks of the canal, along the greater part of the length, have been solidly faced, so as to resist, as much as may be, the disruptive effects of the wash of heavy ships passing rapidly. A local engineer tells me, however, that he doubts whether any big vessel, no matter how powerfully engined, could steam at more than ten knots in the canal, the water being so shallow, and the inevitable result of the attempted rapid motion of a 10,000 ton craft in such a narrow channel being to push a bage volume of water in advance of her, and so to create enormous resistance. But I need not, after all that has of late been written on the subject, dwell further either upon the construction of the canal or upon its commercial future. Other matters claim attention.

One aspect of the importance of the North Sea and Baltic Canal has certainly not yet received the attention which it merits; and that aspect is the strategical one. The commercial advantages of the work are sufficiently patent to be easily distinguished by all; but it would be a mistake to imagine that the practical German government has been induced merely, or even mainly, by commercial considerations to spend its millions on the new waterway. If commercial considerations had been the only ones to be kept in view, nothing is more certain than that the whole affair would have been left to private initiative, and that the imperial government would have vouchsafed very little material support to the venture. But, in fact, commercial considerations are very subsidiary ones in comparison with the strategic gains which the rulers of the new German empire have sought for and have at length attained. In order to fully understand what these are, one must briefly survey the strategical situation from the naval point of view of some European countries which, up to the present moment, have shared with Germany the peculiar disadvantages from the influence of which she has now relieved herself.

These countries are France and Russia. Each of them, like Germany, has hitherto possessed a divided naval force, capable of concentration only with the assent, or, failing that, in the event of the inability to prevent it, of certain other powers. The whole force of France at sea may be accepted as about equal to three times the whole naval force of Germany, or as about equal to twice the whole naval force of Russia. This being so, the relative naval strength of the three great military powers of Europe, as estimated by the amount of materiel belonging to each, may be roughly expressed by the following figures: France, 60; Russia, 30; Germany, 20. The accuracy is at least amply sufficient for purposes of illustration. France normally maintains about two-thirds of her naval force in, or dependent on, the Atlantic ports on her seaboar

two portions subject to the good will or to the impotency of the powers holding the roundabout sea routes, by which alone these two portions could reach each other.

One division, valued at 10, had its headquarters at Wilhelmshaven, where it was subject to observation, if not blockade, by a French force valued at 20; the other, also valued at 10, had its headquarters at Kiel, where it was subject to observation or blockade by a Russian force valued at 20. Even if Russia were not hostile, there would still be risk of a French force, valued at 20, taking up such a position off the Scaw as to be able to prevent the Kiel and Wilhelmshaven forces from joining, and to defeat each in detail, should it venture out. But the North Sea and Baltic Canal has altered all that. Whereas formerly Germany could not hope to meet either France or Russia upon equal terms, she may now feel pretty confident of being able to oppose equal forces to either. Moreover, she has secured for herself the interior position. The way, by sea, from the mouth of the Elbe, at Brunsbuttel, to Kiel Bay, in the Baltic, occupies about 62 hours, at a speed of 10 knots.

By the canal the passage, at a speed of only 5 knots, can be made in 12 hours; so that henceforth the two divisions of the German fleet can unite in less than one-fifth of the time needed for the union of hostile forces, observing the Elbe, on the one hand, and Kiel on the other; and, in addition, they can unite without any possible interference on the part of the foe It is not too much to say, then, that the existence of the canal doubles the strategical strength of the German navy, so far, at least, as it may be called for for employment either in the North Sea or in the Baltic. The cost has been, roughly, £7,800,000. To have actually doubled the German fleet would have cost, at the lowest computation, £14,000,000.

It is true that the canal although it has been spoken of as uniting Kiel and Wilhelmshaven, does not literally unite them; for Wilhelmshaven is not in the mouth of the Elbe

be regarded as, for all practical purposes, a German roadstead.

The strategical importance of Heligoland to Germany was not realized, and was, indeed, laughed at in England at the time of the cession; but a full appreciation of the services to which, in case of need, the Baltic Canal is to be put throws a new light upon the subject, and vindicates the carefulness and foresight of the German government. An English Heligoland might have gone far toward entirely neutralizing the value of the canal; a German Heligoland gives it a singular completeness.—W. Laird Clowes, in the Nineteenth Century.

THE TIN PLATE INDUSTRY IN THE UNITED STATES.*

UNITED STATES.*

In regard to the questions of price and consumption of tin plate, it is thought that the present price will decline considerably, perhaps as much as 25 per cent, but that this will result in an increased consumption, new uses being found for the cheapened material, as, for instance, in packing many articles of food now packed in wood or glass. A reduction in price, how ever, would almost certainly be accompanied by a corresponding reduction in wages to reduce the cost of manufacture.

As to the financial results, it is certain that the most successful manufacturers will be those who can get the cost of production down to the minimum without impairing the quality of the product. The manager of one of the large works states that with careful and economical management and a full product of the plant, there should be a fair, legitimate profit even at present prices, but that the erroneous idea that the business is a veritable gold mine will be dispelled before many of the new works have been running for a year. He considers that a carefully conducted plant ought to net about 20c.—10d.—per box, but that a much higher figure cannot be rensonably or safely estimated.

Another matter which I have made the subject of

ought to net about 20c.—10d.—per box. but that a much higher figure cannot be reasonably or safely estimated.

Another matter which I have made the subject of inquiry is the employment of Welsh workmen, and the results show a very decided diversity of opinion. One company which is making extensive enlargements of its works writes as follows:

"Regarding the nationality of employes, doubtless, many of them will be Welsh. This question, however, is never taken into account, and we would as soon have a good American as the most expert Welshman, for we find the American workmen fully as capable in anything they undertake to do."

Another firm, which does not roll its own black plates, says:

"We are entirely independent of Welsh workers, and have none in our factory."

Two other firms say that they do not find it necessary to employ Welsh workers in any of their departments, and still another firm, commenting on this matter, say that labor saving machinery does away with much skilled laborer, one of the large companies writes as follows:

"We find our most successful workmen to be young Americans who have had some experience in rolling mill work, and that these, then properly directed, very quickly equal, if they do not excel, the Welsh workers we have employed."

On the other hand, another large company writes as follows, but evidently makes its statements too broad in regard to other manufacturers, as will be seen by comparing the following remarks with those preceding:

"In answer to your inquiry as to whether we find it necessars we to employ Welsh workers in all the depart.

On the other hand, another large company writes as follows, but evidently makes its statements too broad in regard to other manufacturers, as will be seen by comparing the following remarks with those preceding:

"In answer to your inquiry as to whether we find it necessary to employ Welsh workers in all the departments of our plant, we will say that we do, and so will any American manufacturer. If you will stop a moment and consider the matter, you will readily understand that the timplate industry is a new one in this country, and that the American workmen have to learn it in all its branches. For this reason it is necessary to have Welsh workmen here to teach the Americans how to make tin plate. When the silk industry was introduced into America from France, French workmen were brought to this country to instruct American workmen and teach them the business. And still further back Frenchmen went to China and learned the business there. So you will see that any new industry started in any country will have to be learned at first from experienced workmen in that business."

The above does not sound very convincing, and the industry certainly seems to be old enough to no longer be dependent upon foreign workers, and this view appears to be borne out by the other statements quoted. In regard to these diverse views I give the following communication sent to me by a Welshman in this country who is thoroughly familiar with the development and condition of the American industry:

"As to the advisability or necessity of employing Welsh workers, I am not surprised to learn that in this connection you find a diversity of opinion among the manufacturers. This difference of opinion is due more to the conduct of those Welsh workers who have already come to this country and are employed at the tin plate works than to any other reason. There is no question as to their ability, and, of course, for the time being their superiority as "skilled workmen in this branch. A large number of American workmen employed in the industry,

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which the iron and steel workers or black mill men were enrolled a reduction of 30 per cent. in wages of rollers, 25 per cent, for heaters and about 30 per cent, for the other branches of labor. The men in the organization refused to accept the reductions and serious labor troubles began. Several mills were shut down, while others—mainly those in the natural gas belt of Indiana—continued with non-union labor, working more or less under difficulties from the hostility of the union men. After several conferences between the representatives of the manufacturers and the representatives of the labor organization, a compromise scale was agreed upon in January, and is to continue in force until June 30, 1805, when the question will again be taken up. Under this compromise scale the men agreed to reductions of 15 per cent, in wages for the rollers and 12½ per cent, in wages for the doublers, heaters and shearers, these reductions being for the No. 29 and No. 30 gages upon which the wages are based, while for gages No. 27 and No. 28 the reductions were 12½ and 10 per cent. A number of firms accepted the compromise and signed the scale. Under the old and new scales the wages per ton of 2,240 pounds are as shown by table No. 6, gages No. 29 and No. 30 being taken as the basis, and the wages getting lower with the smaller numbers and higher with the larger numbers of gage, the limits in each case being shown by the table. The "extras" quoted are unchanged, remaining as under the old scale.

TABLE NO. 6.—WAGES IN AMERICAN TIN PLATE

TABLE NO. 6.—WAGES IN AMERICAN TIN PLATE

	25E E E E E E E E E E E E E E E E E E E	
Men. No. 22 and 30. New scale: Dols. s. d.	Gages. No. 8 to 11. Dols. s. d.	No. 38. Dols. s. d.
Rollers 4.50 = 18 9 Doublers 2.36 = 9 10 Heaters 2.17 = 9 0 Sheavers 1.54 = 6 5 Old scale:	2.04 = 4 4 .96 = 3 10 .85 = 8 6 .99 = 4 1	6.46 = 21 1 3.26 = 13 7 8.13 = 13 0 9.21 = 9 8
Bollers 5.30 = 22 1 Doublers 2.70 = 11 3 Heaters 2.48 = 10 4 Shearers 1.76 = 7 8	2.04 = 4 4 .93 = 3 10 .85 = 3 6 .99 = 4 1	7.60 = 90 10 $8.72 = 15 6$ $8.58 = 14 11$ $2.52 = 10 6$

Nors.—(1) The above prices are for steel plate, and for iron plate the ages are 13 per cent, less, except for the shearers.

(3) The above prices for shearers are for shearing on jaw or crocodile shears of job-work on squaring shears. Shearers on modern squaring shears refer §1.—4s. 24.—per turn, but for gage No. 25 and lighter gages 87; cents—72.—per turn is paid. This clause applies only to shearing tin plate. Extras.—(4) Twenty per cent, added for changed steel and 20 per cent, on 15 Seventeen per cent, added for pickle finished large and trans.

on. en per cent. added for pickle finished ison and steel, excep

Ner.) For all sheets sheared into circles in tin and black plate mills where loss exceeds 10 per cent., 20 per cent. extra is paid.) All sheets cut down to smaller sizes in the and black plate mills are

chesis exceeds 10 per cent. 20 per cont. extra is paid.

(6) All sheets cut down to smaller sizes in the and black plate mills are paid for at scale prices.

In regard to this question of wages the president of one of the largest tin plate unanufacturing companies in the United States said that the passage of the new tariff bill of 1894 compelled his company to make a reduction in the wages of the rollers, doublers and heaters. Under the tariff of 1891 the company paid these menwages 150 per cent. higher than in Wales.

The reductions above noted apply only to the menin the black plate mills. The wages in the tinning depastments differ at the various works. In the tinning and finishing departments on Morewood sets at some works the tinman and washman are paid 12 cents—6d.—per box, and at others 11 cents—5½d.—per box. The riser is paid 4 cents—2d.—per box and the assorter 3½ to 4 cents—1¾d. to 2d.—per box. On patent machines the work is usually done by the day, and dippers get from \$2 to \$2.50—8s. 4d. to 10s. 5d.—per day; risers \$1.50—6s. 3d.—per day, and assorters about 3 cents—1½d.—per box.

Practically all the black plate is of Bessemer steel; but the use of open hearth steel has been advocated for certain purposes, especially for plate for deep stamping. The main difficulty in operating an open hearth steel plant can hardly be economical with a capacity of less than 150 to 200 tons per day, as it is necessary to keep the blooming mill constantly employed. As the largest tin plate works would not require more than 125 tons of steel per day, the open hearth plant is ould be operated with a fair degree of economy by casting only small ingots, of a size suitable for the plant con department of the question unless its product is sold in the steel market, or unless several tin plate works. As an alternative it is suggested that an open hearth plant is ould be operated with a fair degree of economy by casting only small ingots, of a size suitable for the bar mill with which all large plants are quipped; and while t

025 to 035; silicon, a trace only. It is said by a steel maker, however, that much tin plate, stamped ware, etc., is made from steel containing 008 to 012 per cent. of phosphorus.

Among the best equipped tin plate plants in this country is that of the Irondale Works, occupying a site of eight acres adjoining the P. C. C. and St. L. Railway, from which side tracks are laid to the mill. A sketch plan of the plant is shown in Fig. 2. There are six hot mills and six cold mills, together with the necessary doubling and squaring shears, heating furnaces, annealing furnaces, picklers, and a tin house with eight tinning sets.

Only three of these sets are running at present, for the reason that the company secured a verylarge order for black plates when it first went into operation, and therefore needed only to make its tinning capacity equal to the surplus of black plate that it turned out. The general arrangement of the plant is in the form of a U, and the raw material comes in on one side, moves gradually round from stage to stage of its manufacture, and is shipped as completed product on the other side. The main building of the rolling mill is a steel structure 200 ft. long and 150 ft. wide.

Four of the hot mills run across this building, and are driven by one of Totten and Hogg's "Etna" engines, of 600 horse power, with cylinders 26 in by 48 in., the engine being in the middle, with two mills on

either side. Steam is supplied by a battery of five tubular boilers, each 5½ ft. diameter and 16 ft. to 18 ft. long. The other two hot mills and the six cold mills are placed in line lengthwise of the building, all being driven by a 600 horse power engine with Corliss valves, supplied by another battery of five boilers, similar to the above. These hot mills are said to be among the heaviest ever used in tin plate work, the rolls being 24 in. diameter, with 19 in. necks, and the housings weighing 22,000 lb. each. There are three squaring shears to take care of the output of all six mills. The Gray pickling machine is of the most improved pattern, improved upon the Welsh machine, and has a capacity of 1,000 boxes per day. The two annealing furnaces have capacities of 30 tons and 20 tons respectively. The tinning sets have Thomas and White pots. The cold rolls are to be put in a separate building and driven by a separate engine, their places being taken by two more hot mills. The object of this is to remove the cold rolls away from the particles of dust, dirt, cinder, scale and sand that are always flying about more or less in a rolling mill, and which have a tendency to destroy the finish of the plate. The plan also enables the cold rolls to be run independently, which is at times a considerable advantage.

The general manager, Mr. Jackson, says that the capacity of the six mill plant is about 10,000 tons per aunum. Good mill practice would require that there should be a loss of not over 20 per cent. from the steel bar to the finished black plate ready for the tinning pot, including both picklings. About 12 to 15 per cent. of the loss is at the squaring and doubling shears, and if the steel is properly rolled the loss of weight through both picklings and the annealing should not exceed 3 per cent. In using acid of 60° B., the two picklings should be done with not more than 5 lb. of acid to 100 lb. of steel. The output of two mills for one week was 155,455 lb. of black iron and 24,617 lb. of white iron making a to

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BOILERS

can be made with a minimum of wasters, the tinning process not being intricate. The tinning sets produce fifty boxes per turn of ten hours, on a basis of 113 sheets of 14 in, by 20 in, plates to the box. In making the I. C. cokes stand ard quality, the aim is to produce a marketable plate, fully up to the standard, at not more than 2 lb, and 4 oz. or 5 oz. of metal. Pyrometers are used at the tin pots to enable a uniform temperature of the grease and metal to be maintained. Experience has shown that there is no money to be saved by buying an inferior grade of palm oil. The branning and polishing will be done by machines when the tin house is finished, but at the present time this work is done by hand, as in view of the experimental work, and the constant improvement of such machines it is considered desirable to delay purchasing until such times as machines are absolutely needed. The endeavor is made not to exceed 5 per cent. of wasters in the tin house, and whenever this is exceeded, an investigation is at once instituted as to the cause. The Newcastle Works include a mill building 250 ft. by 110 ft., with steel columns 24 ft. high, corrugated iron sides, and a floor of brick and cast iron plates; the pickling, annealing, and tinning plant is in a brick building 362 ft. by 60 ft. and there is a machine shop 70 ft. by 60 ft. The rolling mill has six mills, with hot and cold or lls 22 in., 23 in. and 24 in. in diameter, driven by an engine with a cylinder 30 in. by 60 in. and having a 40 ton fly wheel. This present plant has a capacity of 35 to 40 tons of black plates per day; but it has been decided to extend the works, and put in ten or twelve more mills, so that the enlarged plant will have a capacity of 100 to 125 tons of black plates per day; but it has been decided to extend the works, and put in ten or twelve more mills, so that the enlarged plant will have a capacity of 100 to 125 tons of black plates per day; but it has been decided to extend the works, and put in the nor will be sufficient to the great

The Meurer Works have increased their capacity from 800,000 lb. to 1,000,000 lb. per month. The additional building is 95 ft. by 50 ft. with steel columns and roof trusses, brick walls, and corrugated iron roof with skylights. The floor is of concrete. The new equipment includes six Phillips' tinning pots, furnaces, etc. The Crescent Works, recently completed, are laid out for twelve mills, but work commenced in May with four hot mills having a capacity of about 7,000 tons per annum. The tinning plant will be put in operation during the summer. The mill building will be 105 ft. by 175 ft., the tin bouse 105 by 188 ft. and the annealing house—containing also the pickling machines—48 ft. by 48 ft. The rolls will be driven by two engines with cylinders 34 in. by 72 in. There will be two electric traveling cranes of 10 tons capacity, and an electric light plant rupplying fifteen are and sixty incandescent lamps.

The Laughlin Works, now under construction, will rank among the largest. The hot mill building will be 70 ft. by 140 ft., with sides 20 ft. high, and having a shed roof 20 ft. by 140 ft. covering the heating furnaces and a shed roof 25 ft. by 140 ft. covering the heating furnaces and a shed roof 25 ft. by 140 ft. covering the annealing furnaces. The cold mill building will be 62 ft. by 140 ft., with sides 24 ft. high, and having a shed roof 25 ft. by 140 ft. covering the annealing furnaces and a shed roof 25 ft. by 140 ft. covering the nanealing furnaces and a shed roof 25 ft. by 140 ft. covering the nanealing furnaces. The cold mill building will be 62 ft. by 140 ft., with sides 24 ft. high, and having a shed roof 25 ft. by 140 ft. over the annealing furnaces. Both these buildings will have traveling cranes. There will also be a boiler house, 40 ft. by 88 ft.; tinning house, 60 ft. by 75 ft.; and packing house, 60 ft. by 50 ft. The hot mill have ten sets of 24 in. rolls—commencing with four sets. The machinery will include an Ætna engine of 100 horse power for the dothill have ten sets of 24 in. rolls—com

in, and 750 boxes of 20 in, by 28 in, per week respectively.

Morton.—Six heating furnaces, two annealing furnaces, one hot mill train with three stands of 24 in, by 32 in, rolls, one cold mill train with three stands of 24 in, by 32 in, rolls, one cold mill train with three stands of 20 in, by 32 in, rolls, annual capacity 5, 250 tons of black plates. One tin plating set and one terne plating set with weekly capacity for 500 boxes of 14 in, by 20 in, terne plates. Pittsburg.—Four heating furnaces, two hot mills, and three cold mills; annual capacity 3,750 tons of black plate. Five sets turning out 1,500 boxes of 14 in, by 20 in, terne plates and three sets turning out 900 boxes of 14 in, by 20 in, terne plates per week.

Monongabela.—Nine heating furnaces, nine hot mills, and six cold mills; annual capacity, 20,000 tons of black plate.

Atlanta.—Four heating furnaces, two hot mills, and three cold mills; four tin plate sets and three terne plate sets, each of 3,000 boxes of 14 in, by 20 in, plates weekly capacity.

Beaver.—Eight heating furnaces, four hot mills, and

Atlanta.—Four heating furnaces, two hot mills, and three cold mills; four tin plate sets and three terne plate sets, each of 3,000 boxes of 14 in, by 20 in. plates weekly capacity.

Beaver.—Eight heating furnaces, four hot mills, and four cold mills, with an annual capacity of 6,000 gross tons of black plate; six tin plate sets with weekly capacity of 350 boxes, 14 in. by 20 in.; and two terne plate sets with capacity of 150 boxes 28 in. by 20 in.

The above are but a few examples of the numerous works recently added to the list of works in operation, and besides these there are a number of works in various stages of construction, while new companies are being organized and new projects reported continually. In addition to the many new works being established, many of the existing works are making extensive additions to their plant, and among the encouraging features of the outlook for the future of this industry are the large business in tin plate machinery, the improvements in such machinery, and the introduction of labor-saving apparatus. This latter enables an economy to be effected by reducing the number of and the pay roll for the workmen required for a given output, and while this is perhaps not very encouraging from the workmen's point of view, it is a necessary result of the exigencies of close competition and labor troubles. Thus, one firm which has been in operation since June, 1893, has never done any hand dipping, but with four machines has a capacity of 800 boxes of plates 14 in. by 20 in. per week of sixty bours. The only skilled workman employed at this plant is the foreman, and he is able to instruct all new hands to work the machines in a short time. The introduction of labor-saving devices and machinery is a marked feature in the development of many branches of industry in the United States. Such devices are of less importance in England, where wages are so low, but one manufacturer points out that even there they are a matter for consideration, as if 25 or 50 cents—1s. or 2s.—per ton can be

EXPERIMENTS WITH LIQUID GAS ENRICHERS.*

By T. STENHOUSE, F.I.C., F.C.S.

DURING the last few years many gas managers, instead of using cannel coal, have raised the quality of the gas obtained from ordinary coal to the required standard by using liquid enrichers. These have been vaporized by steam in various kinds of carburetors, Generally the gas is improved two or three candles in

^{*} From the Journal of the Society of Chemical Industry

this way. There is, however, a great difference in the few published statements respecting the value of the various hydrocarbons commonly in use—especially of carburine and benzol. Probably there is as great a difference between the privately circulated results believed to have been obtained at various gas works where enrichers of this class are employed. Dr. Bunte, for instance, makes the rather wide statement that between 7 and 9 grains of benzol vapor will improve 1 cubic foot of gas between 4 and 5 candles.

This is equal to an improvement of 1 candle in over 30,000 ft. of gas by 1 gal. of benzol. Lighting (February 19, 1895), Mr. Hunt. of Birmingham, stated that 1 gal. of benzol enriches 9,500 ft. 1 candle, and that 1 gal. of carburine improves 2,800 ft. to the same extent.

In an interview with Mr. Wim. Young, of Peebles, a report of which occurs in the fas World for January 26, 1895, this gentleman credited a gallon of benzol with an enrichment value of only 4,500 candles, and carburine with being only one-fourth as effective. These statements are so greatly at variance that a gas manager, wishing to select some enricher of this kind for his gas, must be not a little perploxed and scarcely able to calculate what the cost will be.

Of course, it is quite possible that gases of the same illuminating power may so differ in composition that no particular enricher will have exactly the same improvement effect upon them. This, however, cannot account for the enormoun differences just noticed.

With a vew of ascertaining the relative values of a few enrichers for coal gas, I have recently made some careful experiments with several liquids, but principally with earburine of 6'68's gs. gr., and commercially pure benzol. The gas used was of about 16 candle power and was improved from 1 to 2 candles. The photometer employed was the ordinary 60 in. open bar photometer at the Rochdale gas works. A Methven 2 candle severe replaced the candles, on as to avoid any trotals with the latter during the tests. The gas, after

as much carburine is required to give the same improvement.

Added in this way practically all the liquid is vaporized under the influence of heat and the stream of gas playing over the bottom of the small flask. Even with enrichers containing small quantities of dense hydrocarbons, there is a mere stain left in the flask after an hour's test. Under these circumstances one may safely rely on obtaining the full duty from the liquid under examination, as it is practically all turned into vapor and carried directly into the flame, the illuminating power of which is being determined. As the flame is never more than equal to 18 or 19 candles in these tests, the gas is never saturated with the added vapor, and consequently will have no tendency to deposit anything either in the larger flask or in the short length of tubing leading to the Argand burner. It may here be remarked that at Rochdale, where carburine has been employed as an enricher during the last two or three years, Mr. T. Banbury Ball, engineer and manager of the gas works, has not found that any of the liquid has condensed in any of the siphon boxes attached to the mains, notwithstanding the severe frost of last winter.

Working in the way just described, the following results have been obtained with the respective liquids:

Vapor, and consequently will have no tendency to deposit anything either in the larger flask or in the store ingith of tubing leading to the Argand burner. It may here be remarked that at Roschidale, where the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as an enricher during the purish has been employed as the purish has been employed as an enricher during the same leas but tollowing suggestions given in the SCT was the purish the purish has been employed as an enricher during the same leas work of making purish the body of many anateurs.

A FOLDING CAMERAA.

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A FOLDING CAMERAA.

The uniting the body is work of making the holders, is 25 to early a subtract of making has been employed as the tripod used as a staff. The entire weight, the holders, is 25 to an entire weight, the holders, is 25 to an entire weight, the holders, is 25 to an entire weight, the during the interior construction some brought and the tripod used as a staff. The entire weight, the during the interior construction some purished the construction of the current weight. The current weigh

30,000,000 tons of coal annually coked in Great Britain, if the coking were done in closed ovens, There is, therefore, every probability that this most useful liquid will be available in the near future in greater quantities. It is already a rival of carburine, and may soon supersede it altogether for enriching ordinary gas of 15 or 16 candle power. So long as benzol can be purchased at 1s, per gal., it is, in my opinion, decidedly preferable to carburine, even at the lowest price at which the latter has yet been sold. Volume for volume it is about 2½ times as efficient as a gas enricher, and there is less loss and danger from evaporation during transit. The temperature at which benzol volatilizes is also a convenient one, as ordinary steam heat is all that is required. The amount of benzol vapor which common coal gas can permanently retain, viz., over 50 grains per cubic foot at 0° C., is greater by far than anything required to enrich low quality gas to any reasonable extent. If used in a rational way, there is therefore no need to fear that the added benzol will be condensed and deposited in the gas mains. the gas mains.

A FOLDING CAMERA.

MANY have doubtless begun work in photography as I did, with a very cheap affair made of tin, a small plano-convex lens and the plate holder forming back of box, so a person would have to carry a dark room with him in order to change plates or else be satisfied with one negative a day while out on a tramp.

Besides being limited in capacity it was inconvenient

The plate, I, also shown full size in Fig. 2, is for fastening camera to tripod; two of these are used, so either upright or horizontal pictures may be taken. Fig. 3 shows full size of tripod head, the body of which is made of heavy sheet brass and of the pattern shown in Fig. 4.

The wings, or side pieces, W, 1 inch wide, are bent down at dotted line forming a triangle, and edges soldered together and a cap. J, fitted over the end.

The clip, K, and plates, I, are used instead of the usual screw, as it is more convenient, for this camera has to be removed from tripod when plate is changed. The bolt, L, holds both the clip and cap in place, but allows the former to turn about in any direction. A screw could be readily substituted for the clip and used with any camera.

The small carriage bolts, M, hold the legs of tripod, which are of ½ inch oak, ¾ inch wide at top and four feet long. The diameter of head complete is 1½ inches.

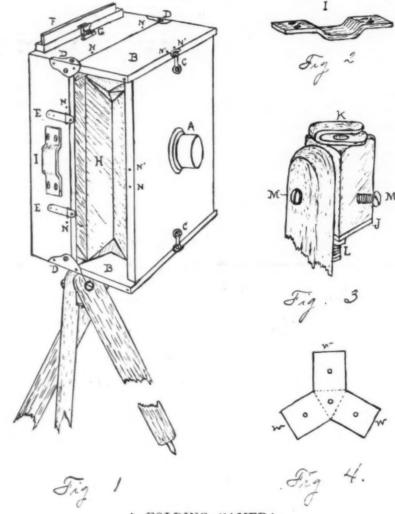
The tripod is opened by a twisting motion and is

four feet long. The diameter of head complete is 1½ inches.

The tripod is opened by a twisting motion and is easily adapted to any position.

As there is no ground glass or finder in this camera, small pins are fixed at certain points, N, N' (Fig. 1), approximately in line with edges of plate and center of lens, and by sighting across them the desired view can be correctly placed on the plate. To a person having facilities for doing such work the cost of the whole outfit, aside from time and labor, would be less than one dollar.

Perhaps a better way of constructing the head would be to have a casting made and drill and tap



A FOLDING CAMERA

the effect of a screen placed before the sensitive This action, which is a purely physical one, present interest, since the same result may be obyusing suitable screens. We know that out this physical action a few coloring substances and at times in very weak proportions have the yof increasing the sensitiveness of the silver recrtain colors without diminishing in a notamer their general sensitiveness.

The sensitiveness imparted by these coloring need does not correspond to the radiations about the coloring matter, but to adjacent radiations also the coloring matter, but to adjacent radiations also the sensitiveness. For example the of crythrosine J shows a band of absorption due-green of the spectrum, while the sensitizet which it produces manifests itself in the of the yellow and the commencement of the This sensitizing effect corresponds sensibly to crum of absorption of the argentic combinacythrosine J.

The production of the argentic combinacythrosine J.

The production of the argentic combinacythrosine J. ble ma sorbe

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The coincidence of the sensitizing effect and of the spectrum of absorption of the argentic combination of the coloring matter is not absolute, and this is made clear if we remark that the coloring matter also acts physically, as a screen, in coloring the substratum of the silver salt. This physical effect tends to counterbalance the chemical effect, and, according to the concentration of the coloring solution, these two actions have a very variable result.

We reach then this first practical and interesting conclusion:

tions have a very variable result.

We reach then this first practical and interesting conclusion:

It is important to select as sensitizers the coloring substances whose effect manifests itself when used in very small quantities.

These bodies alone produce added effects. In making their experiments with orthochromatism the divers operators up to the present time have only used the commercial coloring substances; now these substances represent but a small portion of the organic bodies possessing coloring properties.

The commercial colors are those which combine the qualities required in dyeing; that is to say, solidity when exposed to the light, cheapness, and the property of fixing themselves with or without a mordant on the different woolen, silk, or cotton textiles. These sought for conditions, which are only met with in the substances found in trade, have no value from a photographic point of view, and in a general way the substances that are not used in dyeing, and which possess other properties more suitable for the purpose occupying our attention, are much more interesting.

Taking the above considerations as a basis, we have prepared coloring substances hitherto not used, and have found in a great many of them remarkable sensitizing properties; such are, for example, the tarteines, citreines, oxaleines, succineines, etc.; chlorated, bromated, or iodated products arising from the condensation of acids or of organic anhydrates, with resorcine, metamidophenol, or the homologues of these bodies.

It would require too much space to enter here into

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bodies.

It would require too much space to enter here into the details of the experiments with these substances; they prove that the sensitizers indicated up to this day form but a very small minority of the sensitizers that organic bodies can furnish. The field of orthochromatism has been limited because experiments have been confined to the commercial products made for a purpose which has no connection with orthochromatism.

for a purpose which has no connectant the chromatism.

Having at our command documents relating to the study of more than one thousand coloring substances, we have been able to select mixtures of sensitizers corresponding as exactly as possible to the radiations passing through the violet, green and orange screens, whose use is indispensable for obtaining the three plates representing as negatives the image of the elementary rays, yellow, red and blue; and we have prepared three series of photographic plates presenting a sensitiveness as great as possible to these violet, green and orange rays.

II.

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The selection and the superposition of the monochromes have been realized, thanks to the use of a photographic process, with bichromatized mucilages without transfer.

Glue, soluble in cold water, bichromatized, which does not give photographic images with their half-tones when used alone, acquires this property when substances insoluble under certain conditions are added to it.

If we add, for example, to a solution of glue at 10 per cent 5 per cent, of bichromate of aumonia and from 5 to 10 per cent, of emulsified bromide of silver, and spread this preparation in a thin coating on a glass plate, we obtain a sensitized surface to be exposed to the light under the negative to be reproduced. When the exposure is sufficient the plate is washed in cold water, and we obtain in this manner an image scarcely visible, formed by the insoluble mucilage, an image that may be colored with suitable dyes.

The bromide of silver is afterward got rid of by a

mucilage, an image that may be colored with surface dyes.

The bromide of silver is afterward got rid of by a suitable solvent, hyposulphite of soda, for example.

This process gives with the greatest facility prints of all colors, with all the gradations of the tints of the negative. The bromide of silver may be replaced by other insoluble precipitates. With this process it is easy to obtain polychrome tints by using the principle of the method of Messrs. Cros & Ducos du Hauron.

For this proceed to obtain successively on one plate three monochrome images, red, yellow and blue, coming from three corresponding negatives, care being taken to isolate each image from the preceding one, by an impervious coating of collodion, for example.

ample.

This easy method enables us, by the use of dyes more or less concentrated, or by a simple removal of the color by means of water, to vary the intensity of the monochromes, to modify, if necessary, the effect of the first three coatings by the addition of a fourth, of a lifth, and even more: moreover, it renders the adjustment very easy and makes possible to transfer to paper the ensemble of these impressions.

One of the first specimens of photographs in colors thus obtained, the specimen which accompanies this communication, shows all the practical advantages

which it is now possible to obtain from a method for so long a time neglected. Lyons, May 8, 1895.

AMUSING PHOTOGRAPHY.

AMUSING PHOTOGRAPHY.

However far a photographer may carry the love of his art, there is no doubt that he would besitate to fall from the top of a ladder in order to prove the rapidity of his apparatus by pointing his objective at himself. And yet the photograph that we submit to our readers seems to be the realization of such an experiment.

Despite all, the skillful operator that it represents, carrying pictures, ladder, etc., along with him in his fall, has not experienced the least uneasiness, not even as much as will certainly be felt by our readers at the sight of the tumble represented.

The mode of operating in this case is very simple. The photographic apparatus being suspended at a few yards from the floor of the room, in such a way as to render the ground glass horizontal (say between the two sides of a double ladder—a combination that permits of easily focusing and putting the plates in place), there is spread upon the floor a sheet of wall paper, about six feet in length by five in width, at the bottom



FIG. 1.—THE TERRIBLE FALL

of which a wainscot has been figured. A ladder, a few pictures, a statuette and a bottle are so arranged as to give an observer the illusion of the wall of a room, that of a dining room for instance. A hammer, some nails, etc., are placed at the proper points. Finally, a 5 by 2½ foot board to which a piece of carpet, a cardboard plate, etc., have been tacked is placed under the foot of a chair, which then seems to rest upon this false floor at right angles with that of the room.

Everything being ready, the operator lies down quietly in the midst of these objects, assumes a frightened expression and waits until the shutter announces to him that he can leave his not very pain-



2.—Camera Mounted at the Top of a Step Ladder, E E, in Order to Permit of Operating Downward. O, objective; P, board support containing an aper-ture for the passage of the objective.

ful position. This evidently is merely an example that our readers will be able to modify and vary at their will.—La Nature.

PHOTOGRAPHY IN NATURAL COLORS.

a water color drawing of an Irish peasant girl wearing a red handkerchief over a blue dress, the warm, somewhat sunburnt flesh tints matching the original drawing with almost faultless fidelity—the original being placed above for comparison; delicately colored Indian china and blue china; a lacquered brass microscope with highly reflecting German silver and copper lacquered finishing; a thin uranium green glass tumbler with a subtile play of green and yellow light, and finally, the visible spectrum.

Such a variety of subjects seems to exhaust, almost, the problems which we can submit to the photographer as tests of the range and power of a method which claims to accomplish color photography. The general interest to know how such results can be obtained is not very readily satisfied, for, while the procedure actually involved in securing these photographs in colors is of extreme simplicity—in fact, hardly differing from that with which every photographer is already acquainted—the principles which underlie it need some patient thought to master fully.

Those who have followed the recent development of

phardly differing from that with which every photographer is already acquainted—the principles which underlie it need some patient thought to master fully.

Those who have followed the recent development of composite color photography will readily grasp the principles involved. In the practice of composite color photography the theoretical principles are applied through a procedure far too cumbrous to be available. In fact, the problem of obtaining the naturally colored image by these methods was solved only by the intervention of an apparatus which will unite by optical projection three separate images. In the new method no such projection is needed.

We can hold the plate in the hand, and looking at it against the light, see the objects in their original colors, and this by taking a single photographic image in the camera in, it may be said, the usual manner.

The operations involved in the new method are as follows: A transparent glass plate, which on first inspection appears to bear a uniform tint and to possess a somewhat silky texture, is placed in front of the sensitive film and in contact with it, when the latter is exposed in the camera. Examination of this plate with a strong lens or microscope shows that it is not homogeneous, but is closely lined over with fine transparent lines of three different colors succeeding each other regularly over and over again and in close juxtaposition.

The plates shown at the Royal Society were divided to a fineness of 200 lines to the inch. This is not sufficiently fine to obviate in some cases a linear texture visible on near inspection of the picture ultimately obtained. A fineness of 300 lines to the inch practically accomplishes this, as was demonstrated on a photograph of a group of wall flowers.

The plate which has been exposed under this screen is developed in the usual manner. The result obtained we may regard as embodying in the single minutely divided linear image all three separate pictures required in the practice of the methods of composite photography

ing this triple character, the negative differs little in appearance from the ordinary negative, or a positive subsequently obtained, from the ordinary positive or transparency.

The exposure in the camera is of course somewhat longer, for it is evident that whatever principles are employed, only visible light can be utilized in obtaining a photograph in natural colors, and of this a part is stopped by the ruled screen. Hence a well-lit land-scape may take from three to five seconds with fairly open stop and rapid lens.

Of course, neither the negative nor positive so far obtained shows any color. But if now a plate ruled in three tints, which again are chosen according to color vision theory, is correctly applied to the positive, and if we hold the combined glasses to the light, there is obtained the appearance of the original image as a brilliant transparency in natural colors.

The choice of the tints upon the two screens is based upon the now old hypothesis that all our color sensations, of however varied and subtile tints, are referable to the action, single or combined, of but three sensations (fundamental sensations as they are designated), transmitted to the brain by the color sensitive nerves of the retina. Such a hypothesis, if true, implies evidently that any one sensation must be excited by a considerable range of wave lengths, for otherwise that band of light, the spectrum, wherein the several wave lengths composing white light are, as it were, sorted out and arranged according to wave length, would appeal to the eye only in the red, green and violet; intermediate wave lengths, if not competent to excite the primary sensations in the nerves, remaining, of course, invisible.

The sensation which we term yellow, excited at a particular part of the spectrum situated between the red and the green, is, in short, explained as a resultant sensation arising from a simultaneous excitation of both the red and green transmitting nerves of the retina. Carrying this idea still further, physicists have by measur

PHOTOGRAPHY IN NATURAL COLORS.

A LARGE share of interest was displayed at the recent soirce of the Royat Society in the exhibition by Dr. Joly, of Dublin, of some photographic transparencies upon glass plates, representing various objects in their natural colors.

That every range of color and texture could be dealt with according to the new method by which these were obtained was evident upon examination of the subjects portrayed. The portrait of a gentleman is seated on a garden seat showed the flesh tints of the hands and face reproduced with great naturalness.

The straw hat upon his knee, the buff lining partly revealed within, as well as the faint green reflex on the rim where this caught the greenish light reflected from the foliage among which he sat, appeared reproduced with rhe same fidelity and realism which characterizes the image in the camera.

Pansies of brilliant yellow and brown, deep purpleblack, pale blue, snow white and velvety brown grouped in a painted china vase, appeared with equal fidelity in another picture. Other pictures showed an exterior of a red brick building in Trinity College, Dublin, fronted by a lawn with hawthorns, and above the greenish slates a pale blue sky; a reproduction of

repeated over and over again upon the taking screen. Opacity upon the negative being interpreted as transparency upon the positive, it results that a deep red object, for example, will be crossed by transparent lines upon those parts of the positive image which interpret the action of the "red" taking line in the negative image. The green and violet lines, on the other hand, will be all represented in the image as opaque areas, for their action, when the negative is being taken, will be to stop all dark red light reaching the plate, such rays not exciting green or violet sensation.

The operation of placing the ruled cover glass upon the positive is only correctly accomplished when each of the three fundamental colors upon it lies against a linear area which records the selective action of the taking screen for that particular color sensation.

In the case supposed, a red line will cover a clear space, whereas the biue and violet lines will be blocked out. Hence, the final result will be the red coloration of the image. In general, two lines will are, as the green and red to produce yellow, or the violet and green to produce blue. Or, again, a pure white object upon the final picture will, when examined by a lens, show the three lines, red, green and blue, acting with equal brightness. Thus, although neither white, yellow, blue, pink, nor brown, etc., exists upon the covering screen, all these finally appear correctly as they existed in the color of the original object.

The procedure, in fact, is one in which the three "fundamental" colors are impartially supplied by the covering screen; but the previous experience of the sensitive plate during exposure is such as insures the positive plate during exposure is such as insures the positive plate during exposure is such as insures the positive plate during exposure is such as insures the positive plate during exposure is such as insures the positive plate during exposure is such as insures the positive plate during exposure is such as insures the positive plate d

Were such a substance indeed forthcoming, it and not more faithfully reproduce the true colors of

And this leads us to remark that the particular n

And this leads us to remark that the particular nature of this procedure, resulting in a complete independence of the almost inevitable ultimate fading of pigments, is of no small moment, more especially in the scientific registration of color. For it is seen that the color register is really carried in the silver deposit on the negative or positive, which may, with ordinary care in the photographic manipulation, be rendered quite permanent. And a fading of the tints on the covering screen may at any time be made good by applying a fresh screen. Copies, too, of a picture may be multiplied to any extent.

So far as this new departure concerns the amateur, it is to be presumed that the labor of preparing the screens will not fall to him. His part in obtaining a photograph in natural color will consist in exposing an isochromatic dry plate beneath the ruled screen; and subsequently, temporarily or permanently, applying the ruled cover glass. This is an easy operation. Indeed, a very little practice enables one to do this so readily that it is quite possible to run through a series of lantern plates at an exhibition with the aid of but one covering screen, adjusting and temporarily clamping it over each plate before it is put into the lantern.

However, there is every reason to believe that the

ing it over each plate before it is put into the lantern.

However, there is every reason to believe that the commercial production of the screens will not be attended with any difficulties necessitating a high price.

The lines are ruled with colored inks, made up of gum and gelatine mixed in certain proportions, on a gelatine coated plate, and we believe that all difficulties as to the durability of the pens employed and the nature of the inks bave been surmounted.

That this new development will greatly increase the votrries of photography is hardly to be doubted, seeing that by this method a naturally colored image is as readily produced as one in black and white.—London Times.

THE MAXIMUM POSSIBLE EFFICIENCY OF GALVANIC BATTERIES. By HENRY MORTON, Ph. D.

By Henry Morton, Ph.D.

Some recent publications and other intimations indicate that, with the general revival of business, there are likely to be brought before the public various schemes, such as some of those discussed in the writer's recent articles on "Engineering Fallacies," in this magazine, in which electric energy, derived from galvanic batteries, will be relied upon as the source of power. In view of this, it would seem as if a few words on the above subject might be timely, and the writer has therefore ventured to put into shape the appended calculations.

To discuss this question in an exact and numerical manner, it will be necessary to indicate with precision what class of batteries are referred to, and the writer would, therefore, say at the outset that he refers only to those which long experience has proved to be the most efficient in supplying large currents, excluding those of the Leclanché type which yield only feeble currents.

In all the batteries here referred to, there are the

In all the batteries here referred to, there are the

currents.

In all the batteries here referred to, there are the following common features:

First. The energy is derived from the combination of zine with dilute sulphuric acid. Second. The supply of oxygen required for this combination is obtained by the decomposition of water or some other compound in aqueous solution. In other words, the batteries here considered are the Smee, the Daniell, the Grove and the various forms in which chromic acid is the oxygen-supplying substance.

This being premised, we can begin with the following general statement of principles:

First.—The source of energy being the reaction between the metallic zinc and the dilute acid, its amount can be expressed in British thermal units as follows:

Oxidation of zinc, 2.340 B. T. U.; solution of oxide in dilute sulphuric acid, 666 B. T. U.; or, in all, 3,006 B. T. U. as the total energy developed by the union of the zinc and acid. No arrangement of parts or employment of one material or another in other parts of the cell or for other parts of the reaction can add anything to this, but on the other hand, there must always be more or less subtracted from it to meet the demands of the reaction, to say nothing of internal resistance of the solutions, local action, etc.

Second.—In order that the zinc should combine with the acid, the hydrogen, whose place it takes, must be driven out or otherwise taken care of, and this will demand an expenditure of energy, greater or less, but always considerable.

For example, in the Smee battery the hydrogen is simply driven out in bubbles of gas. To do this requires 2,106 B. T. U. for each pound of zinc dissolved. Taking this from 3,006, leaves only 900 B. T. U. as even possibly available from each pound of zinc consumed in a Smee battery, not counting losses coming from local action, resistance, etc.

This difficulty was realized at an early period and was met by supplying oxygen to take up the hydrogen, and so avoid the great loss involved in expelling it.

To supply this oxygen, various substances have been used, but the only ones of practical importance are sulphate of copper, nitric acid and chromic acid. But even with these, more or less energy must be expended in decomposing them and securing their oxygen. The energies involved are:

B. T. U.

Sulphate of copper, Daniell battery. . . 1,887 Nitric acid, Grove or Bunsen battery. . . . 283 6 Chromic acid, Poggendorff battery. . . . 178 5

If these various amounts are subtracted from the maximum thermal value of the zinc in sulphuric acid combination, we will have for the several batteries:

			D. L. U.
Smee battery, as	before	 	900
Daniell battery.			
Grove or Bunsen			
Poggendorff (chr	omic acid)	 	2,827.5

These figures represent the absolute maxima of energy which a pound of zinc could develop in these forms of battery, excluding all losses from resistance, etc. To get a practical view of these results, however, it will be necessary to reduce them to equivalent foot pounds of work and to horse power rates of doing

Joule has shown that each British thermal unit

allowed, as it must be, for the resistance, local action,

allowed, as it must be, an active etc.

It may, however, be asked: If such an improvement has been made as above shown from the Smee battery giving one-third to the Poggendorff yielding 1½ horse power, may we not expect further improvements as great in amount? To this I answer certainly not in this class of batteries. The entire energy of the reaction between the zinc and dilute acid is 3.066 British thermal units. This would represent 2.3:0.632 foot pounds, or 70.62 horse power for one minute, or 1.17, say 1½, horse power for one hour, and this would be an absolute maximum which could never be reached, far less exceeded.

be an absolute maximum which could never be reached, far less exceeded.

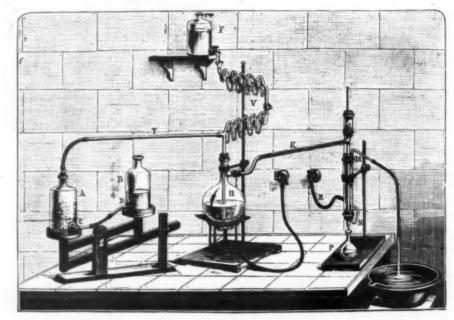
Of course, if we could use some other, and more efficient, reaction than that between zinc and sulphuric acid, some gain might be secured, but nothing of that sort has ever been accomplished, nor from what is known of the combining or thermal equivalents of the available elements, is much to be expected in that line. At all events, we may well accept this as a certain fact, that in any known form of galvanic battery the round figure of a pound of zinc per horse power per hour is an outside figure for efficiency, and when any one asserts that more than this has been secured, there is certainly some mistake or fraud.

ARTIFICIAL ALCOHOL.

The alcohol that we produce and consume is obtained, as it was in remote times, through the distillation of alcoholic liquids. We have improved only the distillatory apparatus and the method of preparing the saccharine material and regulated the fermentation. At bottom, the principle is the same. We always ask Dame Nature to convert the elements into hydrates of carbon, starch and sugar for us by her powerful methods. Our science permits us to finish the rest and appropriate the natural products to our multiple needs.

multiple needs.

Industry, aided by the physical sciences, is asking to be freed from the voke that it has been wearing for centuries; and it is right. The methods employed by



APPARATUS FOR THE PRODUCTION OF ARTIFICIAL ALCOHOL.

equal to 772 foot pounds, and this means that the energy expressed by the heat which will raise 1 lb. of water 1 Fahrenheit (this is the British thermal unit) would lift 1 lb. 772 feet, or 772 lb. 1 foot. If then, we multiply the figures given above, we shall have the various energies expressed in foot pounds of work. In other words, a pound of zine, consumed in these various batteries, would develop the following numbers of foot pounds, all losses from resistance, etc., being excluded:

	Foot pounds.
Smee battery	694,800
Daniell battery,	1,095,468
Grove battery	2, 101, 694
Poggendorff battery	2, 182, 820

If, in each of these batteries, 1 lb. of zinc were consumed in a minute, then the above numbers of foot pounds would represent the work developed in a minute in each case, and to turn this into horse power we should divide each number by 33,000, because a horse power is a rate of doing work of 33,000 foot pounds each minute. This will give us the horse power represented by the solution of 1 lb, of zinc each minute in each hattery. nted by the

Smee battery..... 21 05 H. P. for one minute.
Daniell battery.... 38 19 " " " "
Grove battery ... 68 66 " " " "
Poggendorff b'tt'ry 68 57 " " "

Such a rate of consuming zinc as a pound a minute would, of course, require an immense galvanic battery and indeed it is usual to express the consumption of fuels generally in pounds per hour. To get the horse power due to the consumption of zinc at the rate of a pound an hour, we divide the above figures by 60, and this gives the horse power developed by a pound of zinc consumed during an hour as follows:

This shows that in the best forms of battery an allowance of one horse power for each pound of zinc consumed per hour would be a liberal one, if something is

nature are too slow, and often uncertain in consequence of climatic conditions, fine or bad weather, etc. The manufacturer desires to prepare in a few days or a few hours what time takes months and years to produce. He wishes to have it in his power to prepare at will and at any moment determinate quantities of the substances contained in plants or in the products of animal life.

The conquests of science are already very great in this direction. Colors, perfumes, medicines, sugar, etc., may be prepared artificially by the hand of man alone without the least intervention of the natural forces.

We are now going to show how one has succeeded in preparing artificially, and from minerals, alcohol freed from the injurious principles that ordinarily accompany it and against which hygienists and physicians fulminate.

Chemically considered, alcohol is a compound of carbon hydrogeness.

freed from the injurious principles that ordinarly accompany it and against which hygienists and physicians fulminate.

Chemically considered, alcohol is a compound of carbon, hydrogen and oxygen. These elements are abundantly distributed upon the earth. Wood or cole, water contains hydrogen and oxygen.

Mr. Berthelot was the first to obtain synthetic alcohol, that is to say, alcohol prepared artificially and starting from carbon and hydrogen. Upon causing an electric arc to form between two carbon rods placed in a globe traversed by a current of hydrogen, we obtain acetylene. Upon combining this gas with nascent hydrogen, we obtain ethylene, which, absorbed by concentrated sulphuric acid, is converted into sulphovinic acid, which, diluted with water and raised to ebullition, gives alcohol. This process is too expensive to enter into practice to dethrone grape or potato spirits; but, since electro-chemical science has permitted of obtaining acetylene gas very cheaply, the production of artificial alcohol has been studied anew and simplified. We may recall the fact that this product is obtained by decomposing carburet of calcium with water, the carburet itself being produced by very strongly heating a mixture of lime and coke in an electric furnace. With hydraulic force a ton of carburet of calcium costs about twenty dollars, and a ton of carburet decomposed by water disengages 880 pounds of acetylene. The following are the reactions that permit of very easily producing alcohol with acetylene,

The sulphate of protoxide of ammoniacal chrome absorbs this gas and converts it, especially when hot, into ethylene. The reagent gradually passes to the state of chloride of peroxide. In order to bring it to the state of protoxide and make it capable of absorbing acetylene anew, it is necessary to reduce it by zino or, iron and sulphuric acid. But such reduction is much more economical if nascent hydrogen disengaged by the electrolysis of water be used as the reducing agen. The transforming reagent may remain constantly with the same activity and continuously change an indefinite quantity of acetylene into ethylene. It suffices that the power of the current be sufficient to produce the hydrogen necessary, say a tenth in weight of the quantity of acetylene brought into play. The ethylene, once produced, is absorbed by hot sulphuric acid 69-85° C.) This acid is diluted with water and raised to ebullition. Alcohol is then disengaged. Upon a distilling column being used, alcohol of from 90° to 96° is obtained.

The accompanying figure represents an arrangement that permits of preparing artificial alcohol in a continuous manner, and in hydrogenating the acetylane at the moment of its formation in the very apparatus that produces it. Into the flask, A, is put a misture of carburet of calcium and zine (44 pounds of the first and 5.5 of the second). Into the flask, B, is put water acidulated with sulphuric acid (5 quarts of water and 6 pounds of acid). The two flasks are connected by a rubber tube, CD, and placed upon a support, S, that permits of raising or lowering the flask, B, according as one wishes to increase or decrease the current. The acidulated water attacks the carburet and the zine, and hydrogen and acetylene form; but the two gases immediately combine to form ethylene, which is disengaged through the tube, T. There it passes into the balls of ah Otto apparatus, V, in which it dissolves in contact with the hot sulphuric acid that gradually and constantly flows in from the flask, F. The acid, saturated with

beverages.

Will artificial alcohol, from an economical standpoint, be able to compete with the industrial alcohol
such as now manufactured from potatoes, beets and
grain? Such is the question that presents itself in
the first rank, our alcohol having its diploma for

the first rank, our alcohol having its diploma for hygiene.

The process that we have described permits of obtaining alcohol at about seven cents a quart. This is a splendid enough result, but upon using hydraulic power to produce the carburet of calcium for obtaining the hydrogen, instead of using zinc and acid, this cost would be reduced to six cents. From a practical point of view, the absorbing reagent has been improved and the acid replaced by a salt that is constantly regenerated without the necessity of concentration. The most costly part of the operation has thus been suppressed.

The alcohol under such circumstances costs no more than from three to five cents, at 96° and in the greatest state of purity. As may be seen, the electric industries have in reserve for us quite an encouraging future. Nothing would be more curious than to see the manufacture of alcohol and brandies transported to the mountains, and the enormous power that is now lost, and awaits the desired moment to produce effective work, thus utilized.—La Nature.

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M THE NINETEENTH CENTURY.] RECENT SCIENCE.

I. ARGON.

I. ARGON.

No substance in nature seemed to be better known to chemists than atmospheric air. The composition of air taken from the most different localities and altitudes had so often been analyzed by the best chemists and physicists that up to the last few years it seemed almost inadmissible that any gas existing in the atmosphere should have escaped detection. However, modern chemistry disposes of such perfect methods of analysis and our modern laboratories are supplied with such wonderfully precise instruments—it is sufficient to say that in a modern weighing the incertitude is inferior to \$\pi_0 \text{200}\$ part of one ounce—that when the study of air and other gases was again taken in hand with the aid of the new instruments and methods, a vague suspicion began to grow up. "After all," it was said in scientific circles, "atmospheric air is not so very well known," and it possibly may contain small quantities of some unknown gases mixed with its principal components—nitrogen and oxygen, carbonic acid and vapor of water.* These suspicions are now fully confirmed. When the researches of Lord Rayleigh and Prof. Ramsay were published in full, it became evident that atmospheric air contains over one-half per cent. of some gas (or maybe gases) formerly unknown, and that this gas—named argon by its discoverers—is possessed of chemical properties which offer many a puzzle to the chemist. The distrust which the announcement of the discovery was met with in August last has been dissipated since, and the question, What is argon? stands now foremost.

Is it an element which, like hydrogen or oxygen, cannot be decomposed into still simpler bodies—a "chemical individuality," as Mendeleeff says, which

*Mendeleef, in his Principles of Chemisty (English edition, vol. 1, p. 228, note 13), already expressed the opinion that under the electrical discretars the introduced and the chemical discretars the circusary expressed the opinion that under the electrical discretars the circusary expressed the opinion that under the elec

Mendeleef, in his Principles of Chemistry (English edition, vol. i, p. 256, note 13), already expressed the opinion that under the electrical discharge the nitrogen of the air may be partially dissociated, giving origin to monatomic molecules (N. Helimbotts, having received the news of the discovery of a new constituent of the atmosphere, said that he aiways thought "that there as something more in the atmosphere" (Lord Rayleigh's lecture on Argon at the Royal Institution).

maintains its individual character even when it combines with other individualities? Or is argon a mixture of several new elements? Or is it a compound of well known elements which were never met before in that special combination? These questions press themselves upon every one's mind. However, up to the present date they have not been answered, and most probably the answer will not be given for some time to come, not only because the discovery of several other gases, but also because argon is so peculiar in its behavior as to raise a host of questions of paramount importance for chemistry. The general reader, accustomed to get from science ready results, may, therefore, feel disappointed when, after having perused the following pages, he only finds a number of new unsolved problems east upon science. But, to follow step by step the inquiry which is now going on, to share the hopes and doubts of the explorers and thus to be initiated into the mysteries of scientific research itself, and into the methods of discovery of scientific laws, is perhaps even more interesting and certainly much more suggestive than to learn some time later the bare results.

For the last seven years Lord Rayleigh has been en-

For the last seven years Lord Rayleigh has been engaged in remeasuring the desisties of the commonest gases, with all the precision obtainable from modern appliances, and his work was soon recognized to be a standard work. However, even in the earlier stages of his researches, while he dealt with oxygen and air, there appeared certain discrepancies between his otherwise most accurate results, which, precisely because the measurements were so perfect, could not well be explained by unavoidable errors and created a certain uneasiness as to the permanence of the constitution of air. But when he came later on to deal with nitrogen things took a more serious aspect. Nitrogen is an element; and, whether it be obtained from the air or from one of the nitrogen compounds, such as ammonia, it must always be the same gas, endowed with the same physical and chemical properties. And yet this was not the case. Nitrogen obtained from the atmosphere by any one of the usual methods was regularly by about one-half per cent. heavier than nitrogen obtained in the chemical way from some compound. In each of the two sets of determinations the measurements beautifully agreed together; but the two sets totally disagreed, although all possible precautions had evidently been taken to prevent contamination by other gases, and a strong control was exercised to detect contamination if it had taken place. The disaccord had to be explained.*

The nearest explanation was, of course, to find fault with the chemically prepared nitrogen; notwithstanding all precautions, it might still contain some lighter gas—hydrogen, for instance; but test experiments were installed and compelled the rejection of this explanation, so that there remained but one other alternative, namely, that the atmosphere infrogen, supposed to be the purest of the two, was not pure at all; that it contained some heavier gas which enters into the composition of the atmosphere to no small amount, but in some way or another had hitherto complication. The shade of the pure to th

* The average weight of one liter of nitrogen was 1°2572 grammes when was derived from the atmosphere, and 1°2566 grammes for chemically ob-

September 29, 1898, vol. xivi, p. 512. See also his two subsequent com-nuncations to the Royal Society.

munications to the Royal Society.

2 Proceedings of the Royal Society, January 31, 1895; Nature, February 7, 1895, vol. II, pp. 347-396.

3 Mendeleeff, Proceedings of the Russian Chemical and Physical Society, March 2 (14), 1895; and in Nature, vol. II, p. 543; Berthelot in Comptes Rendus, February 4, 1896, tone cax, p. 295 sq.

1 This last, the lithium method, has been experimented upon with success at Nancy, by Guntz (Comptes Rendus, April 8, 1895, vol. cax, p. 777).

course, most wearisome, and even has been made a reproach to Lord Rayleigh and Prof. Ramsay; but chemical bodies must be taken as they are, and those of them which, like argon, refuse to yield to chemical routine are perhaps the most conducive to an extension of chemical knowledge.

Whenever a new body is discovered, its leading physical properties are always the easiest, and therefore the first, to be determined. We thus know about argon that it is a colorless and inodorous gas, having about twenty times (19.7 to 19.9) the density of hydrogen, and much more soluble in water than either oxygen or nitrogen. Accordingly, the air which is dissolved in water contains a larger proportion of argon than free atmospheric air; in unboiled water we drink a greater proportion of dissolved argon than we inhale of it while breathing, and this property may prove of great importance for vegetation if argon enters, as it probably does, into the composition of plants. It requires also a very low temperature for liquefaction: Olszewski, the Cracow professor, whose admirable achievements in the liquefaction of gases have lately been rendered popular in this country by Prof. Dewar, has turned some argon which was sent to him by Prof. Ramsay into a liquid at a temperature of 305° F. below zero, and into a block of opaque ice at 310° below zero.*

achievements in the inqueration or gases nave latery heen rendered popular in this country by Prof. Dewar, has turned some argon which was sent to him by Prof. Ramsay into a liquid at a temperature of 395° F. below zero.*

But these physical properties tell us nothing about what argon is, and all attempts to unveil its chemical nature have hitherto failed. Even Moissan, with his powerful electric furnace, could not overpower its increness. Neither fluorine, which is one of the most active elements, nor titanium, boron, and lithium, which readily combine with nitrogen, could be induced to combine with argon. Berthelot alone, using the silent electrical discharge, has achieved a partial success; he made argon combine with the prof. Ramsay was so small (a little over two cubic inches) that nothing could be said about the produce of combination beyond its being similar in aspect to the products of combination of nitrogen with benzine; † Negative properties are thus all we know about the chemical nature of argon. Even the spectroscope—this precions reconnoitering instrument—is undecided in its indications. The spectrum of argon is quite characteristic. No other known gas or vapor, Mr. Crookes writes, gives a similar spectrum. But when a glass tube fillel with argon is made to glow under the electrical discharge, and the glow is examined with the spectroscope, two different spectra appear—one of them chiefly in the red and the other chiefly in the blue-according to the energy of the discharge. These two spectra may of course indicate that argon is a mixture of two gases, although it is known that nitrogen and other gases also show two spectra under similar conditions; but Olszewski has found that argon has such a definite temperature of liquefaction, as well as such a definite errical temperature and pressure, that, if it has a mixture, the mixture are present knowledge of them entire the electrical temperature and prospers. The such as a mixture of two gases, although it is known that nit of the proportion of archie

The corresponding temperatures for nitrogen are—818° and—858° F. to oxygen, it has not yet been brought into a solid state, but it liquefies—307°. The critical temperature for argon is—186° F., and the critical searer amounts to 50° a timospheres.

at — 397°. The critical temperature for argon is — 189° F., and the critical pressure amounts to 50° d atmospheres.

+ From a subsequent communication of Berthelot we learn that another sample of argon, also prepared by Mr. Ramsay, behaved quite differently from the former. Eighty per cent. of the former combined with benaine, but only six to ten per cent. of the second would enter into the same combination (Comptes Rendus, April 16, 1866, tome cxx, p. 798). Did the former contain so much nitrogen?

‡ For critical temperatures and pressures see a previous "Recent Science" review (Nineteenth Century, April, 1894).

§ By giving an eighth group (or column) to the second series, which is an "even" series—several other even series also having their eighth groups—and by having certain properties characteristic of the eighth groups or columns.

† J. H. Gladstone's letter in Nature, February 21, 1895; and E. A. Hill's "Argon, Prout's Hypothesis, and the Periodic Law," in American Journal of Science, May, 1895, p. 405.

† Comptes Rendus, 1895, tome cxx, p. 364.

nitrogen, which is lighter than argon; consequently, we cannot be sure that its density is exactly 19-9; it may exceed 20, and even approach to 21, in which case its molecular weight would be about 42; and then argon, in all probability, would be nothing but an allotropic form of nitrogen. We know, indeed, that the atmosphere contains a varying proportion of ozone, which is nothing but a condensed form of oxygen grouped in molecules of three atoms each (0-), while the molecule of common oxygen contains only two atoms (0-). It is therefore possible that nitrogen, too, might appear in two forms; with a triatomic molecule (N₀) in argon, and with the usual biatomic molecule (N₀) in ordinary nitrogen. This is the hypothesis toward which Mendeleef, Berthelot, and Professor Dewarindine, and various circumstances yield it a certain support, namely, the concurrent appearance of argon and nitrogen in nature, the difficulty of separating them from each other, their inertness, exaggerated in argon, their common lines in the spectra, their double spectra themselves, and the outer resemblance between their benzine compounds in Berthelot's experiments; perhaps also the fact that a small quantity of argon was found in nitrogen obtained from one of its compounds.*

However, certain measurement relative to the heat absorbing capacity of argon—too technical to be discussed in this place—seem to point out that, under our present conceptions as to the arrangement of atoms in molecules, we ought to consider the molecule of argon (like the molecule of meroury vapors) as consisting of one atom only. In this case the weights of both its molecule and its atom would be equal to 40. But not only is there no room for such a body in the periodical system—the place being already occupied—but argon would stand by its chemical inertness as a unique exception in a classification which indicates the chemical properties of every other element from its position in the system. The periodic law will certainly not be thrown overboard to suit this

II. HELIUM.

the appearance of some epoch-making work on the structure of matter?†

H. HELIUM.

The researches were at this point in March last, when another far-reaching discovery was announced by Professor Ramasy. It being known that most metals and minerals absorb various gases which can be extracted from the metal or mineral; it was natural to inquire whether some minerals might not contain argon. This was done, and in the course of his investigations Professor Ramsay was brought to extract and to analyze the gas which is contained in a lately discovered mineral, cleveite, and which was said to be nitrogen. This gas contained a revelation. It proved to be argon, as Mr. Ramany expected, but argon mixed with some other gas; this gas on spectroscopic examination displayed, among very many other lines, one bright yellow line which at once caught the attention of the explorers. It was not the well known yellow line of sodium, but was identified by Mr. Crookes as another line frequently seen in the spectrum of the sun's chrom caphere, but never obtained before from any terrestrial object. This line, being very characteristic of the gases of the sun's atmosphere, was ascribed several years ago to some element unknown on the earth, but wil ely spread on the sun, which was therefore named helium. Now, this element was finally captured in a glass tube in the laboratory.

One can easily imagine the sens ation produced by the announcement of this discovery. Many chemists had for years searched for helium among the substances which exist on the earth and in meteorites fallen from the celestial spaces, but in vain; while now the longed-for yellow line glittered in the spectroscope, quite unexpectedly, originating from a by-produce discovered in the search for argon! Upon the reception of the welcome news, the Upsala Professor Cleve (in whose honor Nordenskjold had named the unineral at once extracted the new gase ya and Thalen, one of the best spectroscopists of our time, fully confirmed Mr. Crookes' statement. The gas obtained at Up

mosphere.* At the same date we learned from Professor Ramsay that, while boiling eleveite in weak sulphuric acid, he obtained not only the gas supposed to be helium, but also argon devoid of some gas which is usually found in atmospheric argon, and which may be the cause of the high density of the latter. Three or four distinct gases have thus been discovered—or rather preliminarily pointed out by the spectroscope—while several more are already in view.

We thus stand on the threshold of most important discoveries which are sure to throw much light on the



THE AFRICAN TRAVELER, ADOLF COUNT

chemical processes going on on the surfaces of celestial bodies, and certainly will endow the physics of the sun and the stars with important generalizations; while on the other side the discovery of one or perhaps two gases, possessed of low atomic weights, which have hither to been vainly sought for, will undoubtedly free our chemical classification from an incertitude which has prevailed till now. And, finally, the theoretical questions arising from the properties of argon, and

even from the very errors which may have been made during the earlier hypothetical period of discussion, are sure to launch physics and chemistry in a new domain of philosophical speculation. This mass of discoveries, rapidly following each other, may seem be wildering; but they were not unexpected. For years chemistry had cautiously perfected its methods, and minutely accumulated new data in a limited circle of facts. Now, the fruits of that laborious work are rapidly ripening. "Are we facing a new period in chemistry?" Cleve exclaimed at the end of a letter in which he announced his discovery. Undoubtedly we are entering a period when both our knowledge of facts and our theoretical views in chemistry will be immensely wildened.

(To be continued.)

(To be continued.)

COUNT VON GOETZEN'S JOURNEY THROUGH CENTRAL AFRICA.

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CENTRAL AFRICA.

COUNT VON GOETZEN started in December, 1898, on a bold journey through Central Africa from east to west, and when he returned about a year later his geographical discoveries at once made a great name for him in the scientific world; so that all are interested to know something of his life. He was born May 12,1896, at Castle Sharfeneck, in Silesia, and attended the Schnepfenthal school in Thuringia. Later he studied law. In 1887 he was made second lieutenant of the second Uhlan Guard, to which he still belongs. When he was sent, in 1890, to join the embassy at Rome, he conceived the idea of a hunting expedition to K Mimandscharo, and in the summer of 1891 he went, accompanied by Dr. Ebrhardt, from Tanga to the Umba River, along Mts. Usambara and Ugueno to Moschi, and then turned back toward Pangani. Soon after his return Count von Goetzen was ordered to the military academy, but he could not endure a long sojourn is Europe, and in the autumn of 1898 he went to Africa again, this time accompanied by Assessor von Prittwitz and Gaffron, also from Silesia and lieutenant of the reserve of the second Uhlan Guard, and Dr. Hermann Kersting. The object of the expedition was to explore the watershed between the Congo and the Nile and to obtain information in regard to the active volcanoes in Central Africa, of which no very definite reports had reached us. They went from Pangani over Irango to Usukuma and then stayed for some time at Ushiromba, the station of the "White Fathers," south of Victoria Nyanza, where they had a warm reception and had an opportunity of observing the work of the mission, have pushed farther northward than any of the other tribes that are related to the Zulus.

Like the Massai, they lived by pillage, and finally made a treaty with the Unyanwesi chiefs, with whom they made war on the neighboring tribes, laying the land waste and rendering the commercial highways unsafe. Lieutenant Langheld conquered them several times, and now they give no troubl



WANGONI WARRIORS.

§ Mr. Rameay found that his gas (obtained from a mixime with argon has a density of 5:98, and the same monatomic structure as argon, or at least the same ratio of apecific heats.

§ Proceedings of the Royal Society, April 25, 1866; Nature, May 2, 1866, vol. lik, p. 8.

e, May 10, 1896, vol. lil, p. 56.

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the mountain, four days being consumed in cutting a passable path.

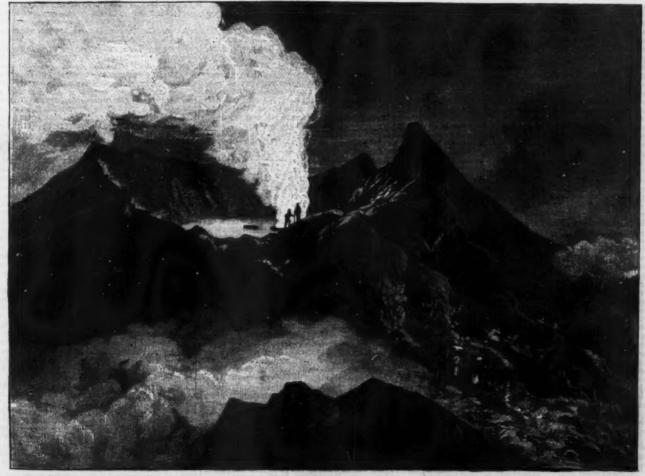
This mountain is shaped like a pyramid with the top out off, and it is overgrown to a height of 9,184 ft, with crica, everlasting and low bushes. The summit of the mountain, which is about 10,990 ft. high, is an immense funnel-like crater about a mile in diameter, with walls 980 ft. high, which are almost perpendicular on the inside, and at the foot of which stretches the smooth cement-like surface of a cooled sea of lava. To quote the words of the traveler, "Standing on the edge of the summer of the crater about 5,000 ft. above the summer of the passable path.

Crater, one overlooks a gigantic arena with steep, high walls; below is a perfectly level floor of a yellowish brown color, and in this floor are two openings, as regular in the pointed cone of Kausumbi. The mighty Kisteria, walls; below is a perfectly level floor of a yellowish brown color, and in this floor are two openings, as regular is under by human hands.

"The mountain, which is about 10,990 ft. high, is an immense funnel-like crater about a mile in diameter, with walls are indeed to the exception of Wihanga, are probably more than 13,100 ft. high. It must have been a tregular intervals, accompanied by a noise like thunsider. On the western edge there is another cruption opening, and later, at a distance of about seven miles, another opening was found from which lava was flow-stretches to the south, from which the Rusisi flows the words of the traveler, "Standing on the edge of the like the pointed cone of Kausumbi. The mighty Kisteria, walls; below is a perfectly level floor of a yellowish brown color, and in this floor are two openings, as regular in the low, picturesque Wihanga with its sunken crater, and the eastern pillar of the Kirunga liquid. The mighty Kisteria, walls; below is a perfectly level floor of a yellowish gali, the low, picturesque Wihanga with its sunken crater, and the pointed cone of Kausumbi. The mighty sunken crater, and the pellowish brown color, and the



FROM COUNT VON GOETZEN'S JOURNEY ACROSS CENTRAL AFRICA: KIRUNGA TSHA GONGO FROM THE LAVA PLAIN, AND THE CAMP OF THE EXPEDITION.—DRAWN BY R. HELLGREWE, FROM A PHOTOGRAPH.



THE PRINCIPAL CRATER OF KIRUNGA.-DRAWS BY R HELIGREWE, FROM SECTIONS AND PHOTOGRAPHS.

level of the sea, and it cannot be much smaller than Lake Albert Edward, for in spite of the clear atmosphere nothing could be seen of the southern and western banks from the northern point, and a very heavy surf broke on the lawa rocks. Lake Kivu is very like the lakes of northern Italy. An archipelago of beautiful little islands, mostly unimbabited, made it easy to go about on the lake in small boats, and as there was no lack of food, the stay on the lake was a real summer holiday for the travelers. But they had to endure the greatest hardship on the march that followed; the region had been pillaged by the slave hunters, the paths were overgrown, and the incessant climbing up and down in the mountainous country was most fatiguing. July 20 they reached the border of the forest, which disappointed the Enropeans, whose expectations had been raised too high by the fanciful accounts of Stanley, but where the swamp was not knee deep the marching was easy, for the underbrush did not grow in the deep shade. On August 17, the expedition started across the charming Lowa, which is very deep and about 330 ft. broad, and on the 21st they reached a great settlement of Wasuaheli and Manyema. On the 37th they camped on the Oso, a beautiful tributary of the river just referred to, which is about 980 ft. broad. From that time to Sept. 10, the date of their arrival at Luvuto, there was great suffering and about thirty died from hunger, exhaustion and poisoning from wild fruits.

But with their arrival at Luvuto all suffering was

But with their arrival at Luvuto all suffering wanded. A Belgian officer came out to meet the expedi But with their arrival at Luvuto all suffering was ended. A Belgian officer came out to meet the expedition, and on Sept. 21 Count Goetzen entered Kirundu, on the Congo, which is often erroneously called Kibonga. It is worthy of note that the objects of the expedition were accomplished in a most peaceable manner, and that the Europeans enjoyed excellent health during the long journey, the result of a careful diet.—Illustrirte Zeitung.

THE RELATION BETWEEN THE MOVEMENTS OF THE EYES AND THE MOVEMENTS OF THE HEAD.*

Delivered before the Oxford University Junior Scientific Club at the University Museum, Oxford, or May 13, 1895.

By A. CRUM BROWN, M.D. Edin., F.R.S., Profe of Chemistry in the University of Edinburgh.

entific Club at the University Museum, Oxford, on May 13, 1895.

By A. CRUM BROWN, M.D. Edin., F.R.S., Professor of Chemistry in the University of Edinburgh.

WE all know that it was a long time before mankind found out that the earth moves. For ages the apparent motion of the heavenly bodies was supposed to be their real motion, the earth being fixed. We who know something of the truth in this matter, do not, however, any more than our ancestors did, see or feel the earth move. We believe that it does so, either because we have been toid by some one who, we think, knows about such things, or because we have reasoned the matter out from data observed by ourselves or reported by credible observers. But in habitual thought and speech we go back to the old assumption, which, for our practical, terrestrial purposes, answers well enough and is perfectly in accordance with our sensations.

When we turn from the great cosmos to the microcosm, when we compare the motion of our own body among the various fixed (terrestrially fixed) and moving bodies around us with the motion of the earth among the stars, we find quite a different state of matters. It never occurs to us that our own body is at rest and that the trees, houses, etc., move. When we really move we not only know but feel and see that we are moving, and every one, learned or ignorant, old or young—if only he is sober—feels and sees that the solid earth is fixed, except on the rare occasion of an earth-quake and in the case of some illusions which we shall have to consider. I wish to discuss the cause of this sensation of the fixedness of the earth, and also incidentally of the exception implied in the words. I have just used, "if only he is sober." If we keep our head fixed and look at any really fixed scene—say, a room in which there is nothing moving, or a landscape, if we can find one, without railway trains, ships, moving beasts, or flying birds—we can allow our eyes to run over it in as uniform or as irregular a way as we please, and see that the scene re

If we have in the field of view a bright object, such an incandescent electric lamp, and after running our wes over the scene before us, shut our eyes, we see

secondary images of the bright object.* Now if the eyes move continuously from one position to another, to the proper that the two secondary images of the bright object corresponding to these two positions a bright band composed of an infinite number of simages each infinitely near its two neighbors. But we were no such band, but a finite number of sharp individual images, each of which corresponds to the position of the eyes during a pause between jerks; unless the bright object is very bright, there is nothing in the two secondary image to represent the positions of the eyes during the jerk.

If for a bright object we take the sun, then we do see bands are fainter than the sharp images and die away sooner. They are the impressions made on the retina by the image of the sun passing rapidly across it during the jerk; but if with the fixed bright object in the tifeld we follow with our eyes a really movime thing, then on shutting the eyes we see a band of light, belicause the image of the bright object passed not very rapidly across the retina. This habit of jerking the eyes from one position of vision to another as fast as the light, well-poised globes can be swung round by the quick-working, straight-fibered muscles which move them may be an imnate habit or it may have pusen acquired by our looking at things and turning it quickly from one object of interest to another; at all development of the property of the pr

of the eye," could lead to a habit, so fixed that we cannot escape it, of moving the eyeballs in the way described.

I have said that the movement of the head, unless it is very rapid, does not affect the fixedness of the glance lines. Translatory motion of our body may be so rapid, as in a railway train, that the eyes cannot twitch so fast as to keep the glance lines fixed relatively to near fixed objects. The eyes do their best; they twitch, but not enough, unless the train is moving slowly, and near objects seem to fly backward. We succeed with fixed objects at a greater distance from us; we can see them fixed, and all objects between us and such visibly fixed objects are seen to move backward, and fixed things beyond them seem to move forward with us. Of course, if by keeping our attention on our carriage and its contents our glance lines become fixed in reference to these really moving things, they seem fixed, and the whole world outside of the carriage is seen to move in the direction opposite to that of our real motion.

It is also obvious that rotation of the head, if it is more rapid than the quickest possible rotation of the eyeball in the head, must affect the position of a glance line, for, in order that the glance line may remain fixed, the eyeball must rotate in reference to the head as fast in one sense as the head rotates in reference to external things in the other sense; but in the case supposed the eyeball cannot do so.

We can try this experiment without having recourse to mechanical means of rotating our body and head, which of course we can do as fast as we please, and a great deal faster than is either pleasant or safe. The most rapid rotation of our head which we are able to produce by the direct action of our muscles is what is known as "wagging"—that is, a rotation about a vertical axis upon the joint between the first two vertebra. In this way it is possible to give the head an angular velocity considerably greater than the maximum angular velocity of the eyeball. When we do this as fast as we can we see that external things do not appear steady. When we wag our head to the right we see the world wag to the left, and vice versa. But the external really fixed things do not appear to us to desertbe nearly so large an angle as the head really does; the eyes make an effort only partially successful, the angle through which external things seem to move being the difference between the actual angular rate of movement of the eyeball in its socket. This difference can best be observed—and, indeed, can be approximately measured—by observing a distant light on a dark night while we wag the head. The point of light seems drawn out into a borizontal line of light, the apparent length of which is the angular difference in question. As we can wag cur head much faster than we can not dit, the apparent length of the vertical line of light into which a bright point is drawn out when we look at it and nod as rapidly as we can is much less than that of the horizontal line of light just spoken of; but I find that I can, by nodding, rotate my head about a right and left axis a little faster than I can rotate my eyes about the same axis, so that the luminous point does appear to be drawn out into a short vertical line. Such violent movements of the head occur sometimes in our ordinary (not experimental) use of our eyes, but they are rare and isolated, so that the disturbance of the fixedness of the grade object on the fixed real will be come

to the clouds or to the moon, and a little practice enables us to change from the one sensation to the other at will.

What has been said seems to show that our immediate sense that the earth and what we call fixed objects on it are fixed is a consequence of the way in which we move our eyes, and in particular, of the way in which, by a suitable movement of the eyeballs, we involuntarily and unconsciously compensate movements of the head, voluntary or involuntary. conscious or unconscious.* That such an immediate sense of the fixedness of external fixed things is of great use to us in moving about among them is plainly shown when we observe the trouble which a drunken man, who has lost this sense, has in guiding himself.

I now turn to the question, What is the cause of this prompt and wonderfully accurate compensatory movement of the eyeballs? There are three sources from which we can obtain information leading to an answer—viz., I, from experiments on ourselves; 2, from anatomical observations and measurements; and 3, from observations of the effects of injuries to the labyrinth of the internal ear. I shall consider these in their order.

By experiments on ourselves I mean the study of the

anatomical observations and measurements; and 3, from observations of the effects of injuries to the labyrinth of the internal ear. I shall consider these in their order.

By experiments on ourselves I mean the study of the effect on the motion of the eyes and on our sense of the fixedness of external things, of movements of our head (in this case always along with the rest of our body), which we do not make, as a rule, for any other purpose. I have already stated that if we shut our eyes, place our fingers on the eyelids, and turn round about a vertical axis, we feel with our fingers the jerking motion of the eyeballs. It, instead of turning round once, we turn round several times—still better, if we seat ourselves on a turning table and get someone else to turn it and us round at a uniform rate—we find that the jerks become less and less frequent, and after two or three turns cease altogether. Another thing which we observe is that, although the turntable is being turned round at a perfectly uniform rate, we feel the rotation becoming slower and slower, and when the jerks of the eyeballs have quite ceased we feel ourselves at rest and have 10 sensation of rotation. Let us for convenience call the sense in which the rotation is still going on positive. This uniform positive rotation has become to us imperceptible (as long as we keep our head in the same position in respect to the vertical), and is what we may call a "new zero of rotation." If the rate of rotation is now increased, we feel this increase as a positive rotation; if it is diminished, we feel the diminution as a negative rotation—a rotation the other way about. What we really perceive then is acceleration of rotation, using the word "acceleration" in its technical sense. If the turntable is stopped, this is a negative acceleration, and what we feel is that we are being turned round in the negative sense, and at the same time we feel our eyeballs have ceased, we open our eyes, we still feel ourselves quite at rest, but we see all external objects tu

world.

It is in reference to this imaginary fixed world that our glance lines are now fixed. If the rate of rotation is changed while the eyes are open, the sensation of rotation is exactly the same as if they were shut; we feel the acceleration—positive or negative—as a rotation in the one or in the other sense, and the jerks of

wink rapout,

+ By "moving the head" I mean me
with the body or any part of it.

1 If we take a sufficiently distant of
may neglect the want of coincidence of
the two eyes, and, moreover, all the
though not are convariantly with one

^{*} I need hardly repeat that by movements of the head I mean movements of the head whether accompanied or not by movements of the body.

the eyeballs take place as if the real external world were not there and we were looking beyond it at the unseen fixed world outside of it—that imaginary world in reference to which our glance lines are now fixed. If, while the experiment I have described is going on, we move so as to change the direction in our head of the axis of rotation—for instance, if after uniform rotation about a vertical axis has gone on, with the head in its usual upright position, until the sense of rotation has ceased, we bow our head forward so that the axis of rotation is now parallel to a line from the occiput to the chin—a very striking and somewhat alarming, but most instructive, sensation is experienced. What we feel is that we are being turned round with a rotation which is the resultant of two rotations of equal angular velocity—one the real rotation about what is now the vertical, the other the imaginary (but equally perceived) rotation in the opposite sense, about the line in the head which was swetical. If the angular movement of the head is small, so that the angle between what is the vertical and what was the vertical is small, then the two component rotations nearly neutralize one another and the strange and alarming resultant is slight; but if the head is bent so that the old and new verticals are at right angles to one another, the real and imaginary components are both felt in full, and the effect is very startling. If the rate of rotation is changed simultaneously with the change of position of the head, we have a resultant of two rotations of different angular velocity.

The most easily observed case of this kind is when the rotation is stopped altogether at the moment of change of position of the head. Here the real component is zero and we have only the imaginary one. This is the case of the well-known practical joke. A man is asked to plant the poker before him on the floor, place his forehead on the end of it, walk round it three times, and then rise and walk to the door. The preliminary part of this experimen

turntable of such a size that one can comfortably lie down upon it.

By the kindness of Messrs Dove, lighthouse engineers, I had the use of a large turntable made for the revolving lantern of a lighthouse. It could be turned round smoothly and uniformly at the moderate speed that is most suitable for experiments of the kind in question. A few experiments with such an apparatus will convince any one that we have here to do with a perfectly definite sense and not with any vague sensations caused by the inertia of the soft parts of the body. This is one of the ways in which the phenomena have been explained by those who hesitate to believe that there can be a definite special sense only discovered within the last few years. That the origin of the sensation is not in the soft parts of the body generally, but in the head, is made perfectly plain by the fact that the position of the head and the changes of that position alone determine the sensations. We must therefore look in the head for the organ of this sense.

In close proximity to the cochlea, which is universally regarded as the organ of hearing, there is an organ

plain by the fact that the position of the head and the changes of that position alone determine the sensations. We must therefore look in the head for the organ of this sense.

In close proximity to the cochlea, which is universally regarded as the organ of hearing, there is an organ of very striking and, I might say, mysterious form. It is found in all vertebrates, and occurs in them fully developed except in the lowest forms of fish. It is contained in a bony—in cartilaginous fishes in a cartilaginous—cavity which communicates in birds and mammals with the cochlea or lagena. This cavity may be divided into the vestibule and the three semicircular canals. The canals open at both ends into the vestibule and each has at one end an enlargement called the ampulla.* Within this bony case is contained a membranous structure consisting of the utricle, situated in the vestibule, and three membranous canals, each in one of the bony canals, each with an ampulla in the bony ampulla and each opening at both ends into the utricle, the saccule, a membranous bag continuous with the cochlear duct, and has in the side next the tympanic cavity a hole in the bony wall filled in by a membrane and known as the fenestra ovalis. The saccule and the utricle have each a spot on the lower wall supplied with nerves which end in air cells known as the macula acustica.

The macula acustica are probably, as suggested by Professor Mach and Dr. Brener, organs fitted to perceive acceleration of translatory motion, and are not connected directly with the function of the semicircular canals. The fenestra ovalis belongs to the organ of hearing, which may thus be said to have a right of way through the vestibule. We need not therefore here consider these organs any further, but confine ourselves to the semicircular canals and the utricle in its relation to them. As already stated, each bony canal contains a membranous canal. The membranous canal is less that the space inside filled with endolymph. The membranous canal is comparatively small. The

perilymph endolymph

pretty firmly attached (in some animals, at all events) to the periosteum of the bony canal. (See diagram.)

STIFIC AMERICAN SUPPLEMENT, No. 1021.

4 Such casal is in all atomas? I have examined, as personimately in ablasm, and if it is important the control property of the control

^{*} in all animals the non-ampullary ends of the superior and the posterior mai have a common opening into the vestibule.

soumou to man and all vertebrate animals, which has remained unknown till about (wenty-two years ago) from the control of the properties of the control of the properties of t





vented, the rebellion will dwindle away in a few

months time.

Convinced of this, the government has sent to the waters of Cuba as many vessels as were found disposable, and proposes to send many more, taking advantage also of the merchant steamers, which will be able to render important help in this exigency.

Our engraving shows some of the vessels which have already been sent to Cuba. They are sixteen in number, of which six are cruisers, one torpedo boat, eight gun boats and one steam launch. Of all these vessels the largest and most powerful is La Reima Mercedes, of 3,000 tons. The new ships Colon, Infanta Isabel and the Conde de Venadito will soon follow. These vessels, together with some others that are already at the Havana station, others that are on the way thither and the armed merchant steamers, are what Spain has to intercept the foreigners who go to the aid of the rebels and who supply them with arms and munitions. Although this fleet appears sufficient for the purpose, in reality it is not, and very soon the government will augment the fleet still further; because the great length of the coasts of Cuba (more than 2,170 miles) facilitates the aims of the enemy, it being very difficult fully to guard so great a coast line. That it will in the end be well protected there can be no doubt; the government knows this to be the most efficacious means of bringing the war to a close.

THE CONDITION OF THE PARTHENON. By SOMERS CLARKE.

By Somers Clarks.

Even those who have not seen the Parthenon feel an interest in the structure; a building which has always held its own as, perhaps, the greatest wonder of refined perfection among all the architectural wonders of the world; and, notwithstanding the grievous blows and buffetings it has undergone, its rains still possess the most extraordinary power of fascination, partly through sheer majesty, beauty and refinement of form and partly from the charm of the material, the color, the situation and associations. That this matchless building stands in imminent danger of ruin, even beyond that which has already befallen it, is but too evident.

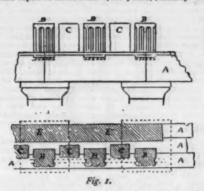
Notwithstanding the explosion of the powder magazine within, which burst out the sides of the cella and overthrew the lateral colonnades, and notwithstanding the bombardments which the western front has undergone, still, from a distance, the building stands up and can be appreciated as a whole, owing chiefly to the fact that the western portice is standing, even including its angle cornices, and that the outline of the pediment is still preserved, although shorn, for the most part, of its cornice.

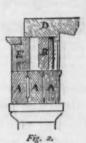
The fall of the angle cornices and their supports would be the entire and final ruin of the structure; a mere skeleton would remain, and it is at those angles that the greatest danger shows itself.

The construction of the building throughout being purely trabeated, there are in the original scheme no thrusts to counterbalance, no arches trying forever to

overthrow their abutments. On the other hand, when stone beams are broken through, they have the trick of acting after the manner of arches, and may often be found to have sunk, until, while the crack gapes at the bottom, the two faces of the break are in violent pressure at the top, and the species of arch thus formed is only kept in its place by the resistance of the adjoining masonry—in fact, thrust is established, and that in a structure where the resistance to thrust forms no part of the original design.

The Pentelle marble, of which the Parthenon is built, was selected with the greatest care, any but the best blocks being rejected. The material is splendidly preserved, but after the lapse of more than 2,000 years, clearly to assist in carrying the great cornice stones,





during which the marble has been exposed in the building, disintegration seems to be, in parts at least, making its way, and once begun, it may be making rapid progress, only perhaps to be made clear to us by the sudden collapse of an architrave under some unwonted although slight strain.

Since casts were taken, at the beginning of this century, of some sculptures still in their place, these sculptures have considerably disintegrated. What is going on with the sculptured surfaces is probably going on eless manifestly in the great blocks of which the Parthenon is built.

The following notes, accompanied by rough sketches and by a reproduction of a photograph, will, it is hoped, make clear, to those acquainted with construction, in how great peril the temple stands.

The sketch diagrams are not to scale, and are only carried as far as is necessary to assist in explaining what has to be said. The beautiful and careful drawings published by Mr. Penrose, in his "Principles of Athenian Architecture," show, much better than I can do, the methods of construction adopted in the Parthenon. But for those who may not have the book at hand, the rough sketches, Figs. 1 and 2, may be of use.

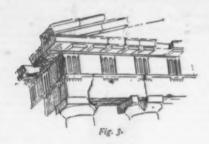
By these it is seen that the architrave is built up of three large slabs, set on edge side by side, and



THE EAST FRONT OF THE PARTHENON .- FROM A RECENT PHOTOGRAPH IN THE BUILDER.

The outer stone of the architrave is cracked through; consequently the great block of the angle triglyph is standing on a very insecure base, and on this block rests the cornice. The abacus and part of the cebinus of the angle capital are gone, while the capital of the next column southward is broken

The outward stone of the architrave has therefore a poor base. In its fall it would almost certainly heave up the work above it—indeed, the open joints of the



masonry at this angle show that there is a slight ten-

masonry at this angle show that there is a slight tendency outward.

Fig. 4 shows the capitals of the fourth and fifth columns of the western portico with the architraves resting upon them. The danger which threatens the outside architrave stones of the third, fourth and fifth inter-columniations cannot be overestimated. The capitals of the fourth and fifth columns have their face completely broken away. In consequence, nearly all the support of the outer slab of the architrave, except a very few inches, is gone.

The vertical joints of the architrave slabs are hopelessly shattered, and there is an ugly crack developing itself in the architrave of the middle inter-columniation. It seems that these stones are kept in their places solely by the weight of the triglyphs, etc., above tying them back, and by the wrought iron cramps with which nearly every block of masonry in the temple is tied to its neighbor.

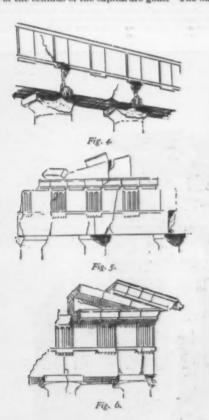
The S.W. angle of the western portico is in equal danger with that at the N.W.

Fig. 5 shows the southeast angle of the east portico. Here we see the laumentable result of the fall of the cornice. The outline of the temple is entirely lost. The cracks and breaks in the masonry make it evident that, although the great mass of the angle cornice has fallen, there is still a considerable tendency to move outward.

The angle shown on Fig. 6 at the northeast corner

outward.

The angle shown on Fig. 6 at the northeast corner of the east portico is perhaps in greater danger than is any other part of the peristyle. Here the abacus and part of the echinus of the capital are gone. The outer



architrave stone is cracked in two places. Upon the outer piece which stands out in the air rests much of the great weight of the triglyph, and over this, much shattered, lie the cornice and the first stones of the pediment. As we look at it we only wonder why it does not full at any moment.

Not only are the outer stones of the architrave broken as above stated, but many of those behind them are in a very precarious state. The insertion of new stones would be not only a thing difficult in itself, but one which would be most fatal to the marvelous effect of color the ruin now presents to us.

Pentelic marble is of a cold sugary white when fresh cut and takes very many years to mellow. It is beyond question that a great deal may be done with bronze tie rods, which would be almost, if not quite, invisible.

We cannot too strongly deprecate anything in the way of "restoration." and the insertion of new stones comes very near to this.—The Builder.

LIFE FROM A PHYSICAL STANDPOINT.* By A. E. DOLBEAR.

By A. R. Dolbear.

I suppose there is no question about which science concerns itself and everybody has more interest in than this one of the nature of life. Some pretend to think we know nothing about it and never can know anything; others are quite as sure that we know it to be correlated with other forms of force and in some way convertible into them; while a third class may hold an agnostic position, content to wait until knowledge shall grow so as to include the nature of life. Still it may be doubted if there be any thoughtful person who does not hold some sort of a theory, about it which he expects will be substantiated, and it is quite certain if any demonstration of the nature of life were to be given to-day there would be a great multitude of persons who would at once declare they had always so held. This expectancy shows itself in so many ways, that one may be sure that nearly every person has some theory of things, some scheme into which he contrives to fit all kinds of facts. That is to say, we can't get along without some sort of philosophy and we make our own if there be none otherwise provided. Even those who pretend to contenn all schematic attempts in knowledge and who mildly reprove such efforts by calling them speculations are easily found to have some pet scheme of their own which finds favor in their eyes.

Now there are speculations and speculations. There

efforts by calling them speculations are easily found to have some pet scheme of their own which finds favor in their eyes.

Now there are speculations and speculations. There is a kind that has been common from the beginning until now, when imagination has full sway with no manner of regard for data or for appropriate facts at all. Such a one was the commonly held view as to the origin of the world and especially of living things, including man. They were created, at their beginning being the same substantially as we see them now to be. There is not and never has been in the history of man any phenomenon that could give warrant to such a hypothesis, yet it has been held and fought for by men now living.

Then there is another kind of speculation that has or tries to have proper data—that shows some respect for experience. Such was the attempt of Robert Chambers in the book called "Vestiges of Creation," a book which is deserving of more praise than I have yet seen awarded it, for he undertook to handle such data as were available to him, and he discerned dimly the process which all naturalists to-day see clearly. His data were inadequate and could not compel belief, but his attempt as compared with the hypothesis it contended against was as daylight to darkness. It had some experience in its favor; the other had none at all.

Lastly, there is a hypothesis derived from the study of groups of appropriate facts, the attempt to find an adequate explanation of all of them, without going beyond the bounds of possible experience, that is, without importing into the phenomena some transcendental conditions. Such is Darwin's Theory of Natural Selection, offered for acceptance as a provisional hypothesis thirty-five years ago. Also fought against stubbornly by naturalists, as well as theologians, in spite of the plain fact that it was either such a hypothesis or nothing; there was no other competing one that had any standing ground at all, which seems to imply that to some minds it is more rational to entertain an unint

to entertain an unintelligible hypothesis with no experimental data in its favor than it is to entertain one that has a considerable body of experimental data for its basis.

Swedenborg taught the nebular hypothesis, but gave no astronomical reasons. Kant developed it, giving philosophical reasons, which were not considered to be adequate. Laplace gave mechanical reasons which were adequate, and he who explains that theory to-day gives the reasons of both Kant and Laplace, but he quite ignores Swedenborg. Kepler explained the orbital movements of the planets as due to guiding spirits. Newton explained them by the doctrine of gravitation and dismissed the spirits from service. In his Principia he says he framed no hypotheses; nevertheless, he was a great framer of hypotheses; as, for instance, the corpuscular theory of light which he worked out, and his theory of a necessary ether which he did not work out. So hypotheses are absolutely needful for guidance in all profitable efforts, and as much so in science as anywhere else. Indeed, what is science if not our correlated experiences? It is interesting to see how men have tried to define it. Buckle says, "Science is a body of generalizations so irrefragably true, that though they may be covered by subsequent generalizations, can never be overthrown by them." Spencer says, "Science is a higher development of common knowledge." Others say, "Science is classified knowledge."

Our experiences of all sorts are the subject matter of science, our interpretation of them is our attempt to be logical, our attempt to be scientific, and a true interpretation of any phenomenon will not be inconsistent with any other truth, that is, it will be consistent with any other truth, that is, it will be consistent with any other truth, that is, it will be consistent with any other truth, that is, it will be consistent with any other truth, that is, it will be consistent with any other truth, that is, it will be consistent with any other truth, that is, it will be consistent with any o

* Biological lectures delivered at the Marine Biological Labor Vood's Holl in the summer session of 1894.

forces, which presided for the time in much the same way as Kepler's guiding spirits presided over

same way as Kepler's guiding spirits presided over planetary motions.

We know what the history of such prepossessions has been. A hundred years ago caloric was thought to be such an imponderable potency, light was thought to be another, electricity still a third. Each of these turned out to be no imponderable at all, but simply physical properties of matter of the ordinary sort. But the change from the old to the new view in these matters made it needful to change the fundamental ideas concerning matter itself.

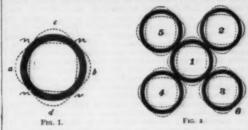
The physiologists for a generation have ceased to think of a vital force as different from other forces in the same way as they have ceased to consider light as an emanation. And the consensus of opinion among biologists, if one may judge from a multitude of expressions by them concerning life, is that all the phenomena exhibited by a living thing are finally resolvable into physical and chemical processes.

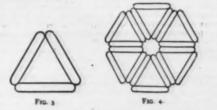
A vital element peculiar to organisms no more exist than does a vital force working independently of matural and material processes.—Claus and Sedgwick. It must not be supposed that the differences between living and not living matter are such as to justify the assumption that the forces at work in the one are different from those to be met with in the other.—Huxley.

Zoology, the science which seeks to arrange and discuss the phenomena of animal life and form as the clause of the law of physics and clauses the phenomena of animal life and form as the chemistry.—Lankaster.

Certain it is that life is a chemical function, says Prof. Stokois, of Amsterdam, and he adds, Is not the chemical function a sort of life?

So vital force, as a distinct somewhat invented to account for living phenomena, has now no status anywhere. If it be so, then it is plain that matter has properties which have not been included in its list. If matter has been defined as inert, or as dead or animalimate, one may have to revise his definition. In a chilbited by living things are to be explained only on the assumptions, first as due to the inherent properties of the matter that exhibits it, or to some external agency—not inherent in it, to which the name vital force is just as good as any? and if this has been discarded for seemingly good reason, then there is the other alternative only. But somehow most men who have thought about lighave felt loath to adopt this. Is not this the same as saying that there is somehow left to be a good reason for refusing to adopt it, even in tiles in the common unanalyzed notion into which we have all been schooled, that matter is deed and inert and out of it can come nothing but so-called inorganic phenomena. Along with this has come a relatively new piece of knowledge called the conservation of energy, which asserts that all the forms of energy are transformable and that the sum of their energies is a constant quantity. None hitherto being able to see how with and physical phenomena are correlated, and o





Just 27, 1896.

SCHNTIFIC AMERICAN SUPPLEMENT, No. 1091.

Affig. ft has it some adjusted at the process of the

quarto, with six plates, on such artificial protoplasm. Quincke found that, if a substance soluble in water be finely powdered and rubbed up with oil and then surrounded with water, the water diffuses into the oil and makes of it a kind of foam, consisting of minute drops of water closely packed together in oil, and thus presents the appearance of honeycomb structure. The soluble substance which works best for this preparation is K.CO. Olein oil is generally used—ordinary oil is useless—and much pains need be taken in preparing it, but when a minute drop of this properly prepared substance is placed in water or a mixture of equal parts of glycerine and water, it becomes clear and transparent and exhibits changes in shape; streaming movements like those of an amoba are seen, which are kept up for hours. It throws out processes and withdraws others, and the drop as a whole will change its position.

Up the center of the processes there is a streaming movement to the end of the process, where it spreads out and flows back in a layer next the surface. The movements are influenced by warmth and by electricity, and one who did not know what it was he was looking at would suppose he was seeing an actual amoba.

Frommann, Klein and many histologists find that

was looking at would suppose he was seeing an actual amoba.

Frommann, Klein and many histologists find that protoplasm consists of a kind of network of less fluid material, the interstices being filled with more fluid material. Indeed, such kind of structure is thought to be true of every kind of animal cell.

This view is an advance from the older view that protoplasm was wholly structureless and homogeneous. Butschil, however, on the basis of his experiments and observations, concludes that protoplasm is an emulsion of two fluids, which mechanically presents the honeycomb structure, and that so far the structure is wholly due to the physical and molecular qualities of the substances which exhibit it, and that what was taken to be a network pecullar to a living mass is really only emulsion. He finds it too from protozoa to vertebrate. The interfibular substance of muscle which has been taken for network by some observers, Butschil finds a honeycomb with transverse partitions, and the fibrated axis cylinder of a nerve has cross strands, indicating this also to be honeycomb. As there are many degrees of fineness possible to such physical structure, it would follow that if there be so-called "structureless" protoplasm, it is only apparently so, because the meshes are too fine to be seen.

there are many degrees of fineness possible to such physical structure, it would follow that if there be so-called "structureless" protoplasm, it is only apparently so, because the meshes are too fine to be seen. The honeycomb structure is believed to be an albumen containing some molecules of a fatty acid not miscible with water; the more fluid parts which fill the interstices are watery fluid containing albumen and alkali. Such chemical substances in such close physical relations would necessarily permit such phenomena of movement as are seen in such microscopic masses of living matter. The shorthand explanation is that these are due to surface tension and chemical actions; so both structure and motions are thus reducible to purely physical and chemical terms. The success that has attended the efforts of chemists in synthetic chemistry has emboldened some of them to assert with confidence their belief that every kind of a combination can be artificially produced, and that when the substance protoplasm is formed it will possess all the qualities of protoplasm, including life. Now albumen, Crieblus Nillous, is very closely related to protoplasm and some kinds seem to be nothing else. Egg albumen contains I sulphur atom for every 70 of carbon, globulin albumen 146 and hæmoglobin albumen 350, or ratios of 1, 2 and 5—a rather striking fact.

Already albuminoid bodies have been artificially

every 70 of carbon, globulin albumen 146 and hæmoglobin albumen 350, or ratios of 1, 2 and 5—a rather striking fact.

Already albuminoid bodies have been artificially made, but they showed no vital qualities. If Butschill's experiments signify anything, they signify that nothing of the sort should be expected from a substance chemically homogeneous like precipitated albumen, for there are required two differently constituted substances, physically mixed, not chemically combined, and no mere chemical process or chemical product could give such a mixture.

It is evident that in a chemically homogeneous mass there can be no occasion for changes of any kind within it, and chemistry alone cannot give us any substance which can give characteristic vital actions.

It is true enough that the materials with which Butschil has made his observations are not the same as the real substance of living protoplasm, yet they are not so far apart as one at first thought might imagine. Whenever chemical action is taking place, whether fast or slow, these exchanges in the form of energy are likewise taking place, changes from molecular to mechanical motions, from one degree of absorption 'and conduction of heat to another, from one degree of condensation to another, and so on, and now let one add to these the quality of atoms, referred to a little way back, namely, that their field of action is not limited to a push or pull by contact, but that it acts at a distance from itself in various ways, and one of these is to compel other masses in its neighborhood to assume the same form and condition as itself—that is, the so-called sympathetic action. It can be apprehended that when there is energy being expended in this kind of a way we have a process which is called growth.

If the molecules are closely adhesive, as they are in excelled salids the growth any take release only a process.

hended that when there is energy being expended in this kind of a way we have a process which is called growth.

If the molecules are closely adhesive, as they are in so-called solids, the growth can take place only upon the outer surface, yet even here the growth is limited to the same kind of unaterial as that of the initiating mass; that is to say, a crystal of salt will only annex salt molecules to itself, so, though there be several different substances in a solution capable of crystallizing, each one will select molecules of its own kind, and each crystal is similar in kind and structure throughout. This is a kind of natural selection inherent in the atoms themselves.

But there is the widest difference in character between the few elements that make up a living thing, from oxygen with communistic instincts to nitrogen with antisocial qualities and strong individual proclivities. If induced or compelled to associate with other elements, it is ready on the slightest provocation to abandon them and become a free rover. Gunpowder, nitroglycerine and the fulminates are examples of the qualities of this element to effect disorganization. This element is always one of the constituents of protoplasm, and one might therefore expect it to be unstable and restless, as indeed it is.

due tikewise to harmonic changes of energy inseparable from the atoms themselves.

Movements that result in change of position of the body are called mechanical; movements that result in the enlargement of the body in one way or another; are sailed growth; movements that result in the enlargement of the body in one way or another; are sailed growth; movements that result in the original of the similarity of the second to the first has been attributed to beredily, a term expressive of a fact, but embodying no explanation. The conditions in the neighborhood of such growing thing, that react upon it in one way or another; the conditions in the neighborhood of such growing thing, that react upon it in one way or another; but in this particular, and this too, has been a bazy term, as applicable to one thing as to another; but in this particular, and the conditions of t

One of the indications of the rate of activity of any kind in an animal is the rate of elimination of nitrogen. This is emphasized here in order to make it plain first that the origin of movement in a living thing is to be traced to the energy embodied in the chemical combinations, and second, that particular movements, or at any rate some of them, which have been attributed to some directing agency—vital force, or life—are due likewise to harmonic changes of energy inseparable from the atoms themselves.

Movements that result in change of position of the body are called mechanical; movements that result in the origin of another similar body are called reproduction—and the similarity of the second to the first has been attributed to heredity, a term expressive of a fact, but embodying no explanation. The conditions in the neighborhood of such growing thing, that react upon it in one way or another, are called its en-vironment; and this, too, has been a hazy term, as

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